

**Surface tension and surfactant
concentration in atmospheric
aerosols:
Completing the
ecosystems-aerosol-cloud relationship**

**Barbara Noziere
CNRS / IRCELYON, France**

Context and scientific objectives

SOLAS Theme 4

“Interconnections between aerosols clouds and ecosystems”

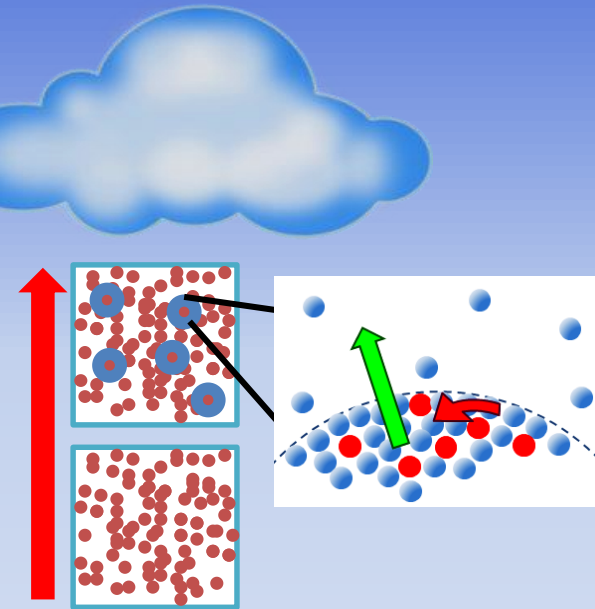
Cloud formation not predicted by models:
computational & fundamental challenges

In Earth's atmosphere clouds possible
ONLY because chemicals in aerosols affect

- Raoult's term $\Rightarrow$ «hygroscopicity»
- Surface tension $\Rightarrow$ surfactants

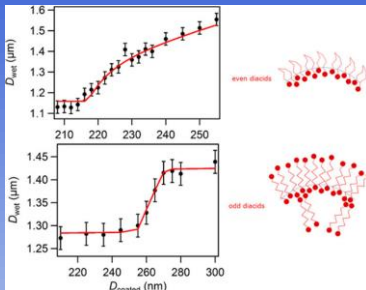
Surfactant effects not detected by «on-line» instruments

$\Rightarrow$ hardly studied and excluded from models



Context and objectives (*continued*)

Yet many recent studies indicate a role of surfactants

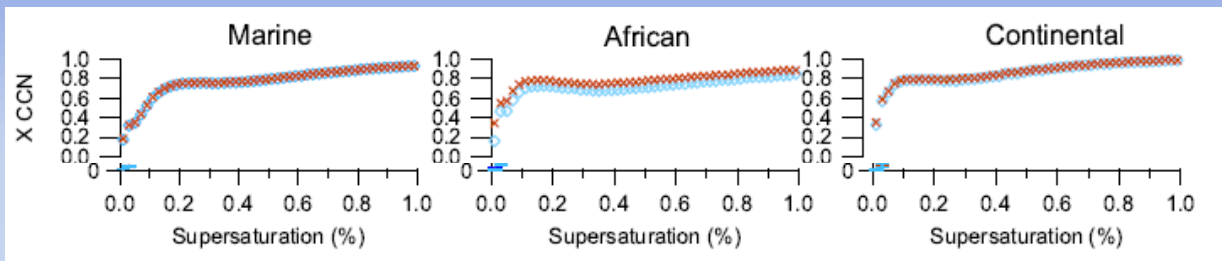


In laboratory

Ruehl et al. 2012; 2014...

In the atmosphere

Asa-Awuku et al. 2006, Good et al., 2010; Irvin et al., 2010



Long res. time instruments $\Rightarrow$ $CCN_{meas} \geq 1000 \text{ cm}^{-3}$ instead of 500
 $\Rightarrow$ surfactants present ($\sigma \sim 50 \text{ mN/m}$) & factor 2 on CCN numbers !!!!

$\Rightarrow$ Objectives of our project

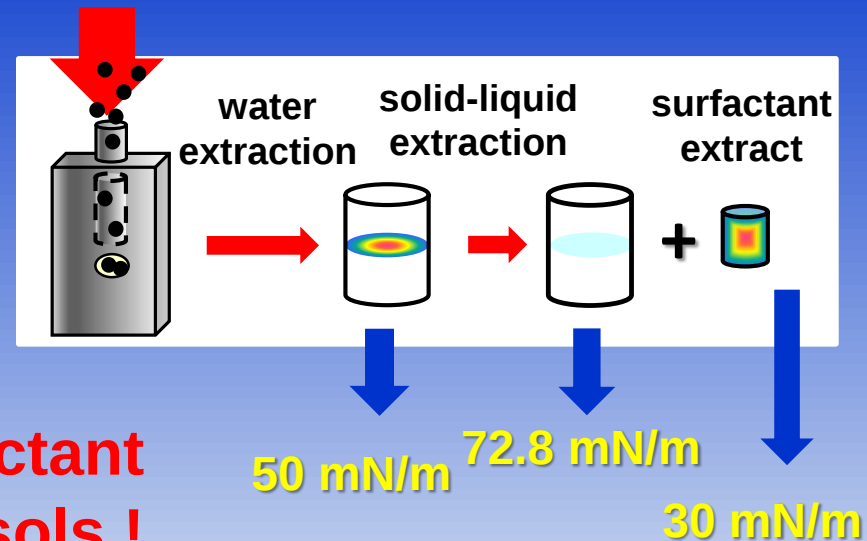
- Isolate & characterize aerosol surfactants directly
- Develop alternatives to “on-line” techniques
- Determine their role on cloud formation
- Determine sources and/or link with ecosystems

Methodology

- Collect aerosol samples on filters
- Isolate aerosol surfactants by targetted double extraction

Ekström et al. 2010, Baduel et al. 2012, Gérard et al., 2015

⇒ **Extracts total surfactant fraction of aerosols !**



Combine with investigation of the extracts by

- Surface tension measurements

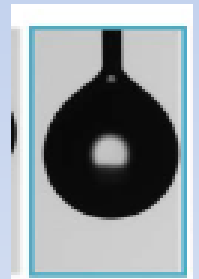
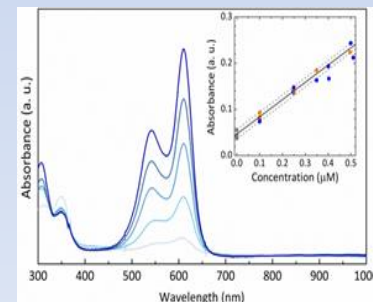
Hanging droplet technique ($V_{\min} = 40 \mu\text{L}$, $\varnothing_{\min} = 1 \text{ mm}$)

- Concentration measurements

Colorimetric methods: dyes specific for anionic, cationic or non-ionic surfactants

- Chemical analysis (NMR, LC/MS...)

...

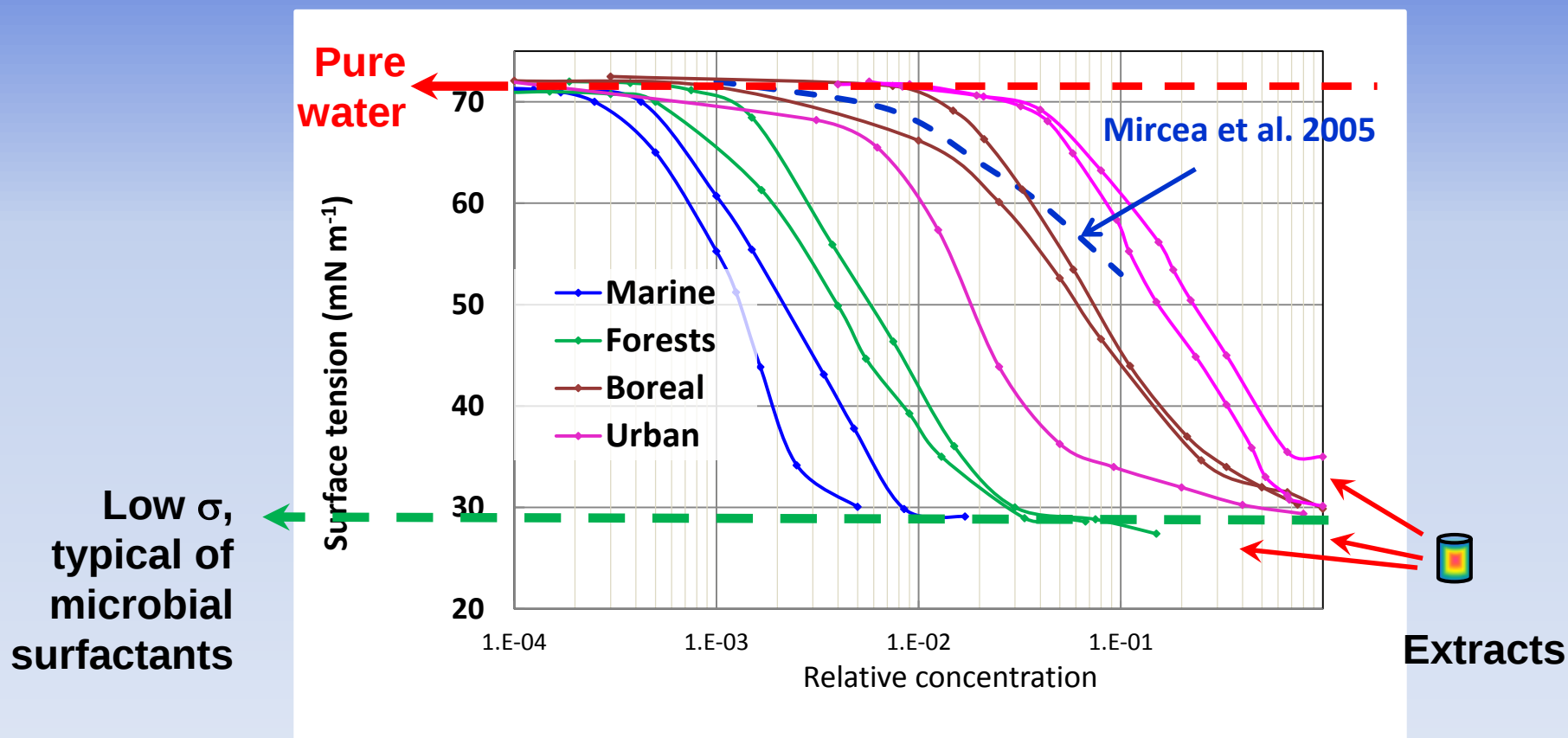


Main results

1. Relative surface tension curves:

Much stronger surfactants than expected present in PM1, PM2.5, PM10 aerosol fractions

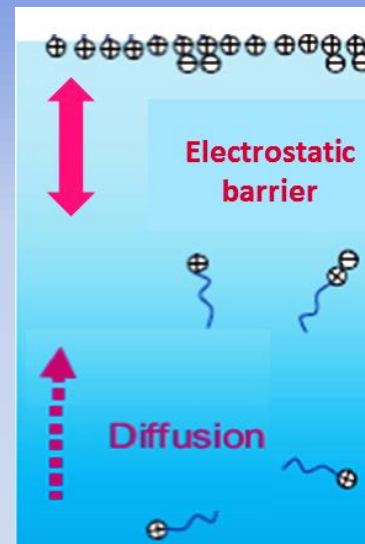
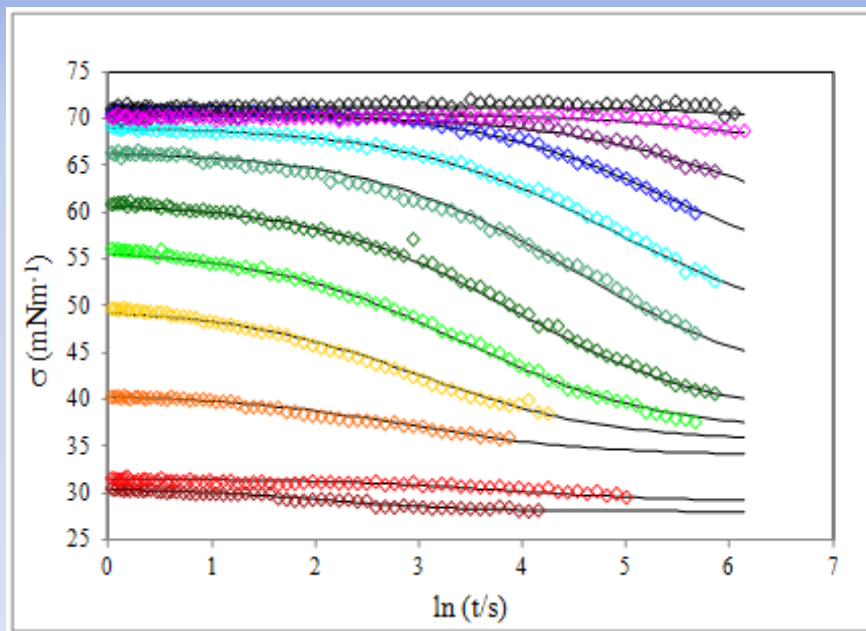
Ekström et al. 2010, Baduel et al. 2012, Gérard et al., 2015



Main results (*continued*)

2. Dynamic surface tension study of atmospheric surfactants:

Atmospheric surfactants need ≥ 30 s to equilibrate in aerosol particles $\Rightarrow$ explains why on-line instruments «blind» to them



nature
COMMUNICATIONS

ARTICLE

Received 2 Aug 2013 | Accepted 28 Jan 2014 | Published 25 Feb 2014

DOI: 10.1038/ncomms4335

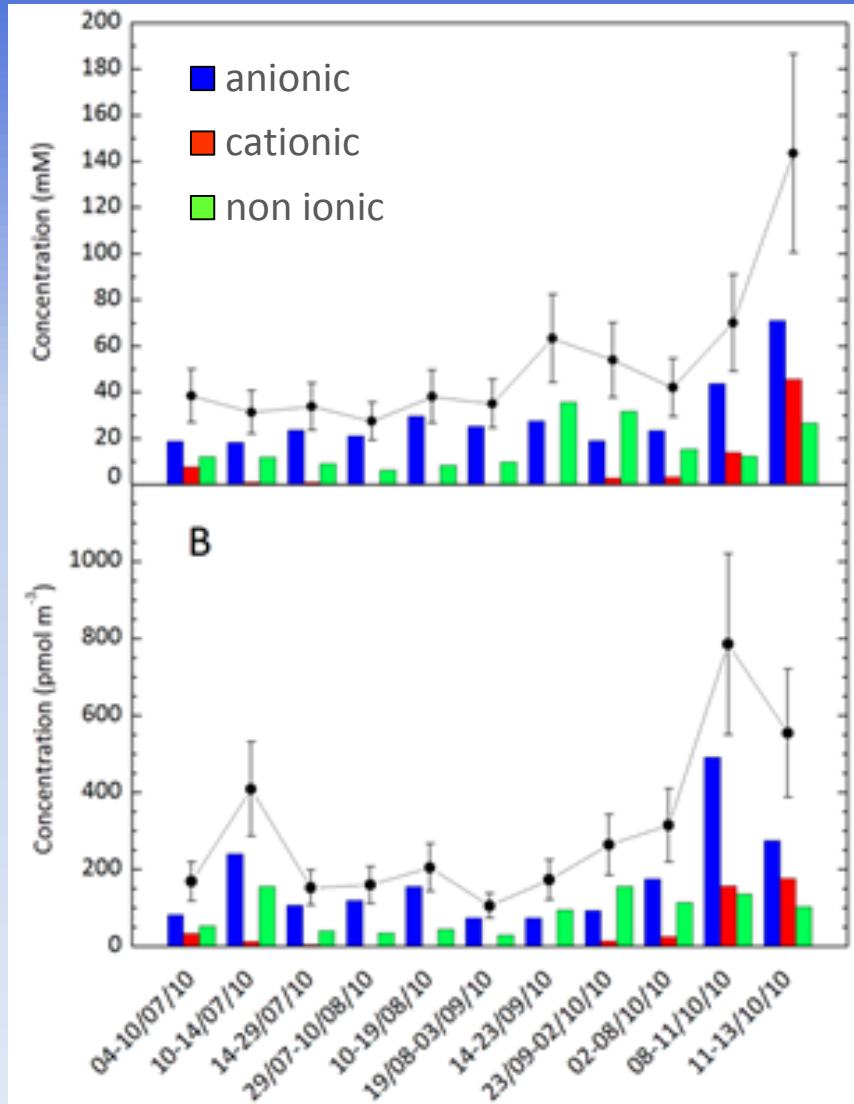
OPEN

The dynamic surface tension of atmospheric aerosol surfactants reveals new aspects of cloud activation

Barbara Nozière^{1,†}, Christine Baduel^{1,†} & Jean-Luc Jaffrezo²

Main results (*continued*)

3. Absolute concentrations & surface tension curves:



PM2.5 from
Askö, Sweden

Seasonal evolution of anionic > non-ionic > cationic surfactants

Detection limits:

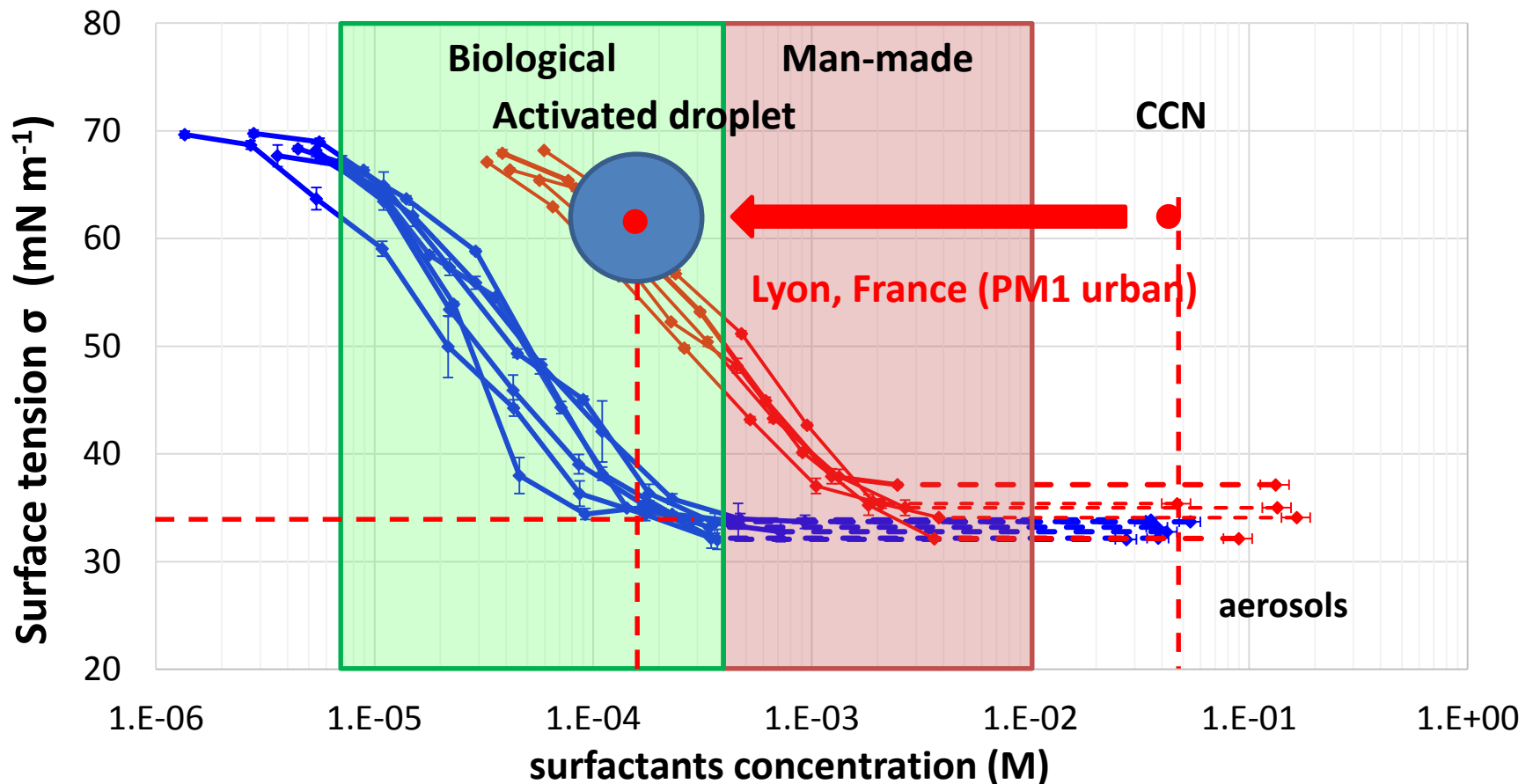
~ 50 nM anionic/cationic,

~ 300 nM non ionic

No correlation between
surfactant types

⇒ no common sources

Absolute surface tension curves

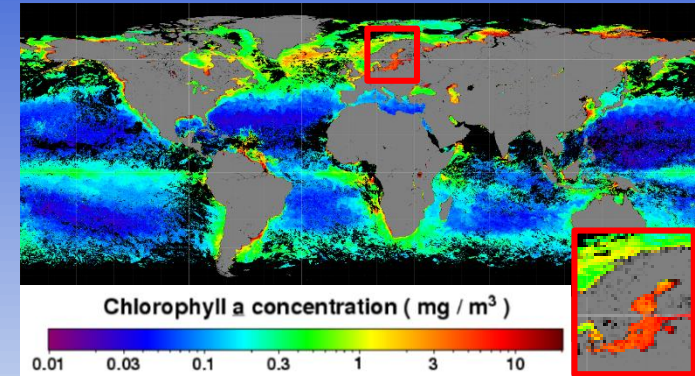


- Surfactants concentrated enough to affect complete activation
- Curves in typical range for microbial surfactants $\Rightarrow$ biological origin ?

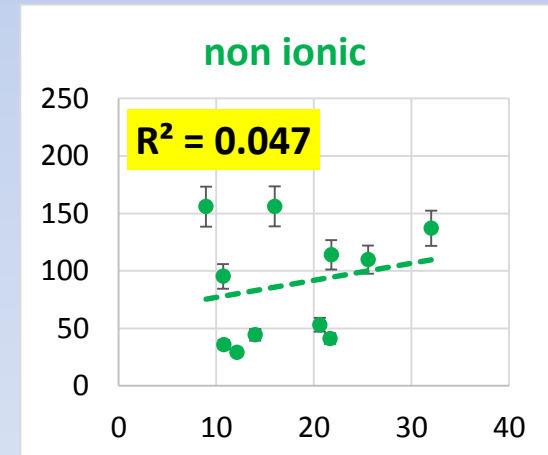
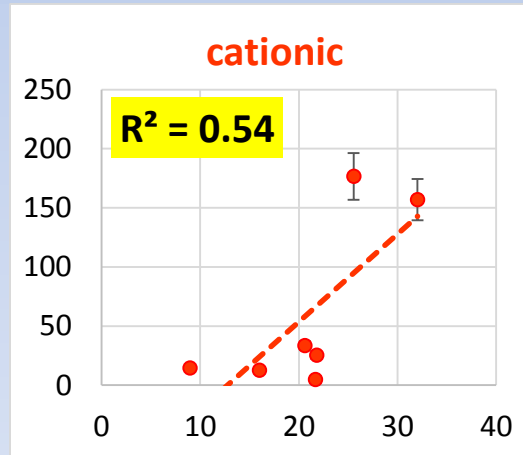
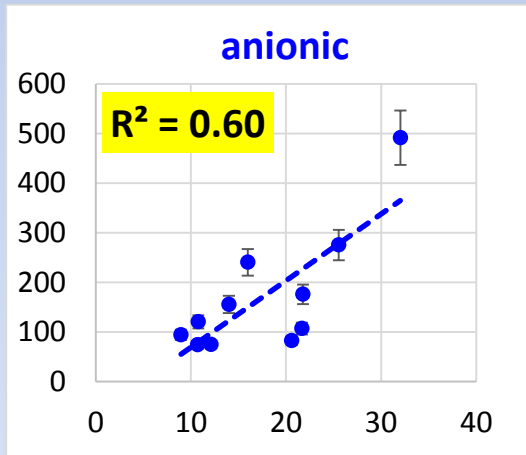
Main results (*continued*)

3. Source apportionment: Correlations with seawater chlorophyll

Chlorophyll a data from
MODIS aqua satellite, Level L3
Resolution: 15 km area around Askö



surfactant a concentration
(pmol m⁻³)



Chlorophyll concentration (mg m⁻³)

⇒ Possible marine & biological origin

(Gerard et al. 2015)

Conclusions

- Presence of strong surfactants ($\sigma \leq 40$ mN/m) in atmospheric aerosols now well established
- Concentrated enough to affect entire cloud droplet activation process
- Now possible to investigate sources and other properties (molecular structure...) $\Rightarrow$ large progress expected in near future
- Development of alternatives to on-line techniques essential for progress

Contributors

V. Gérard (PhD student), CNRS/Ircelyon, 2013 -

C. Baduel (post-doc), Stockholm University, Sweden, 2010-11

S. Ekström (PhD student), Stockholm University, Sweden, 2006-10

Collaborators

R. C. Cohen & A. Frossard, Univ. of California, Berkeley, USA

A.-M. Delort, P. Amato, M. Sancelme, I. Canet, CNRS / Inst. Chimie Clermont Ferrand

E. Asmi, H. Lihavainen, Finnish Meteorological Institute, Finland

S. Frka, B. Gasparovic, Rudjer Boskovic Institute, Croatia

Sponsors

Agence Nationale de la Recherche / National Science Foundation, France-USA project 2013-16

European Commission, International Reintegration Grant 2006-08 + EUSAAR 2009-11

Swedish Research Council, res. grant 2007-08 + FORMAS grant 2010-11



Thank You !