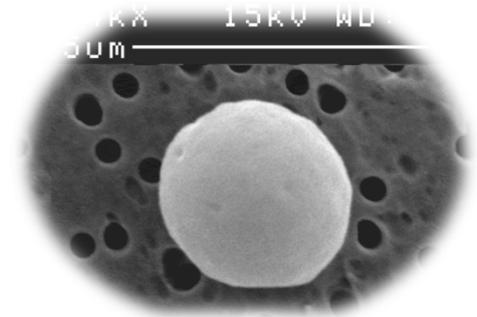


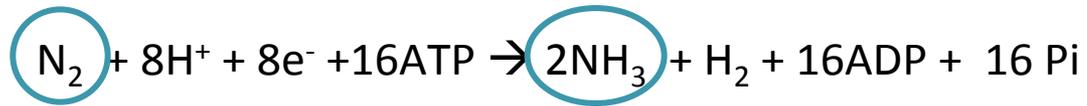
# Impact of atmospheric iron from Saharan dust on *Crocospaera watsonii*

*Violaine Jacq*

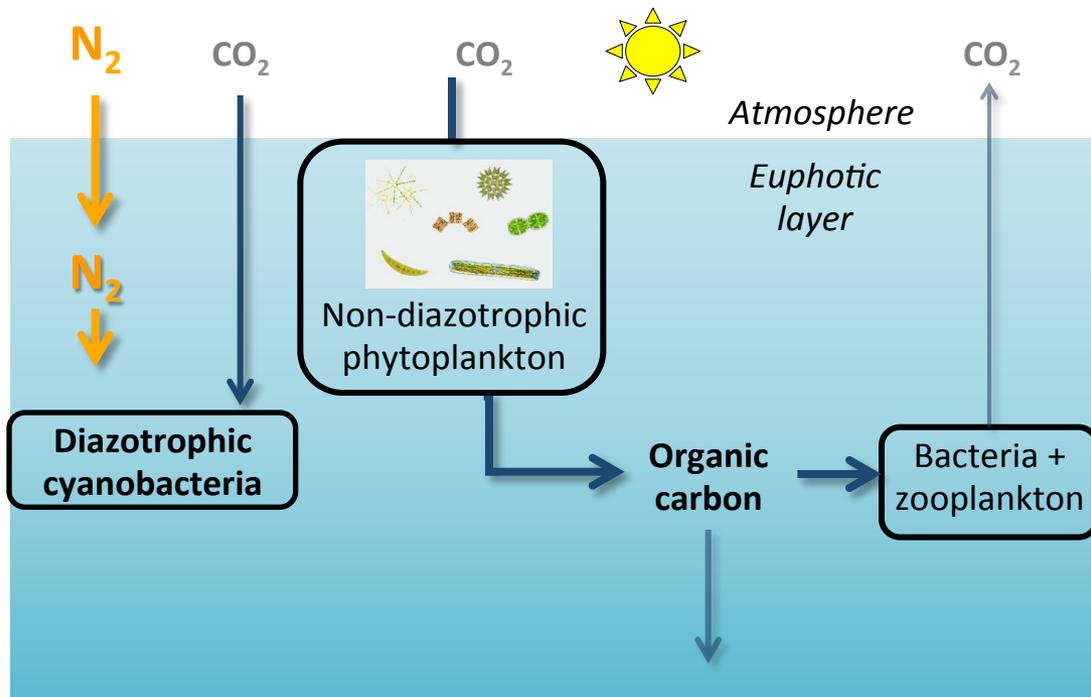
Meeting SOLAS Paris, 29 juin 2015



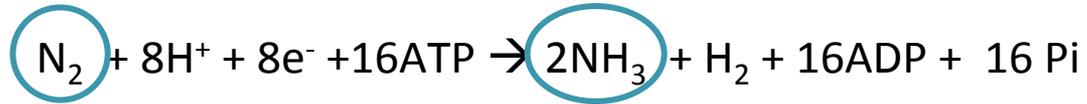
# N<sub>2</sub> fixation



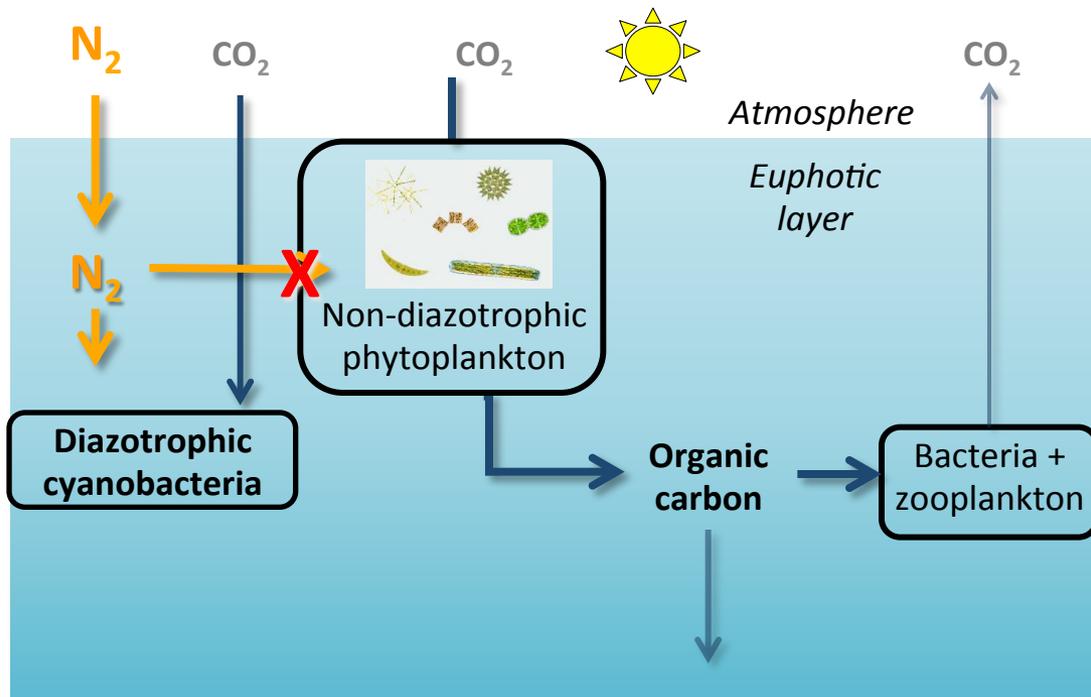
Nitrogenase enzyme complex



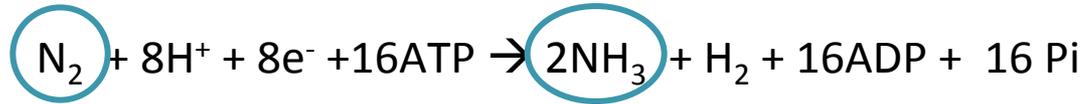
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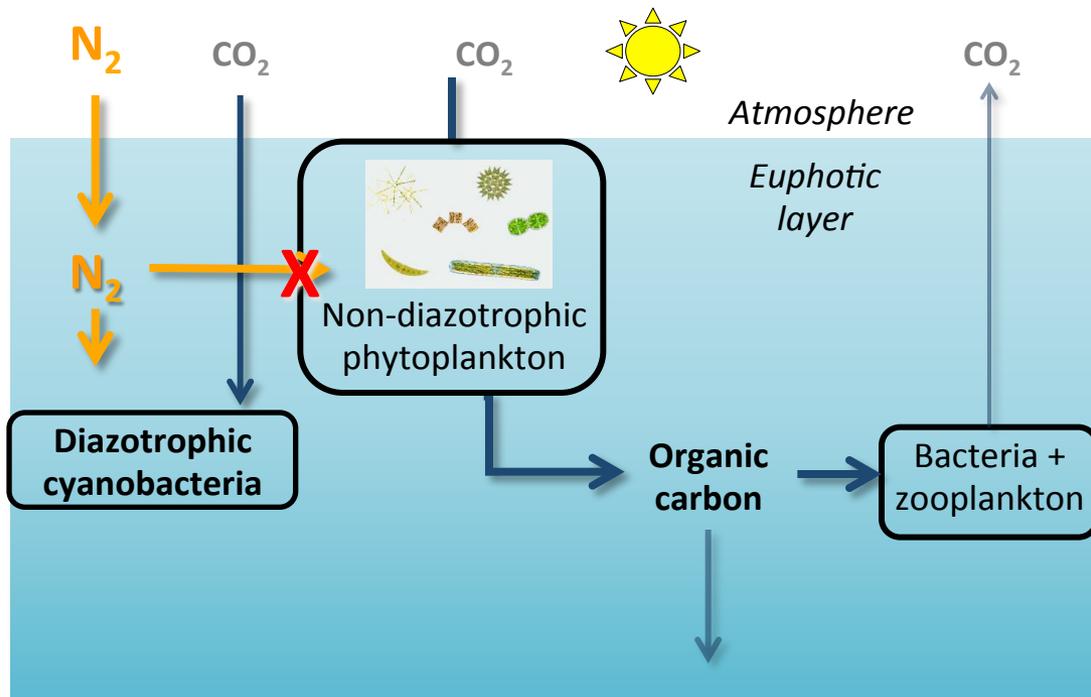
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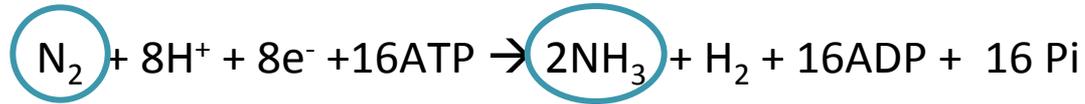


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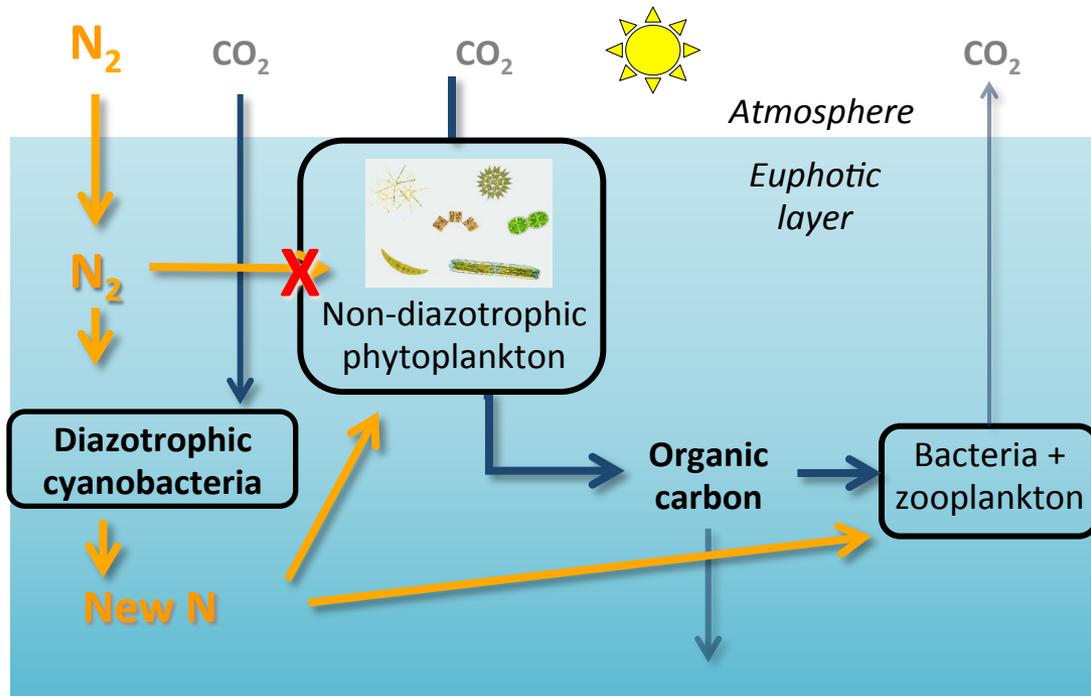


**Ecological advantage of diazotrophic cyanobacteria in N-limited areas**

# N<sub>2</sub> fixation

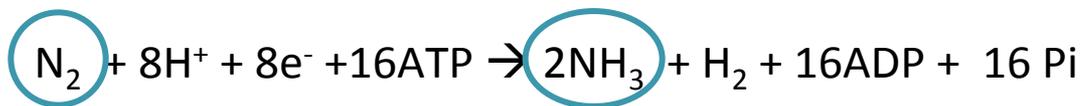


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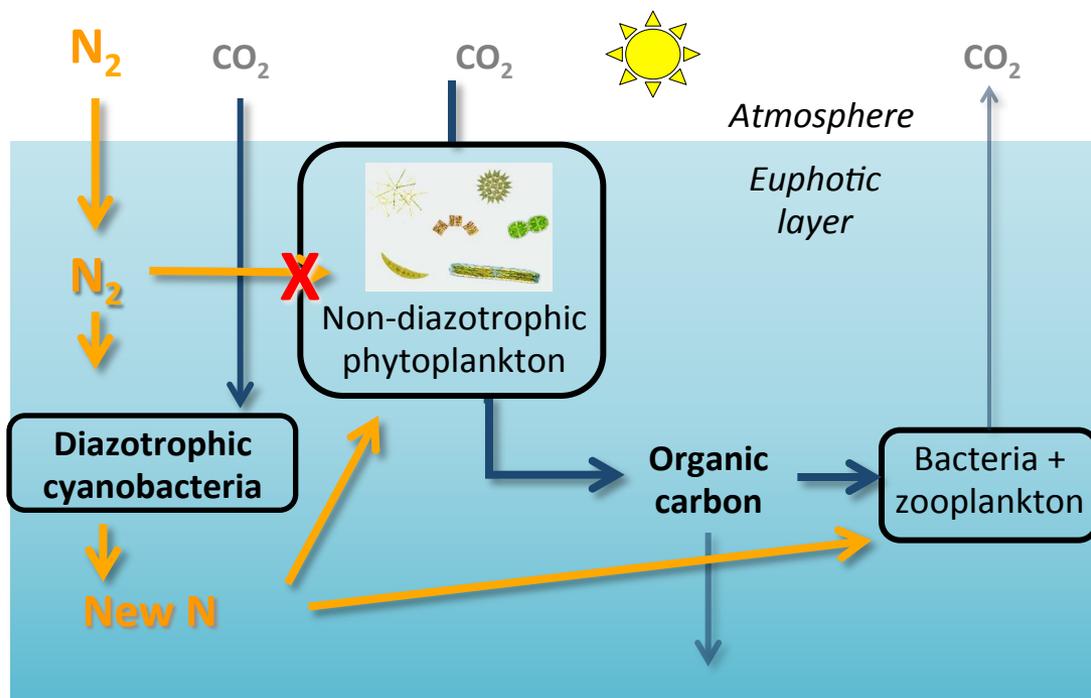


**Ecological advantage of diazotrophic cyanobacteria in N-limited areas**

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Nitrogenase enzyme complex

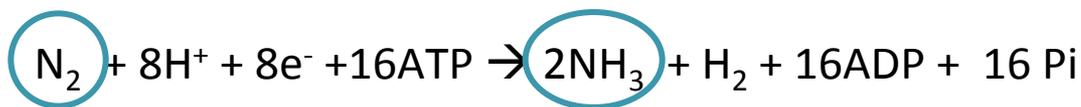


**Ecological advantage** of diazotrophic cyanobacteria in **N-limited areas**

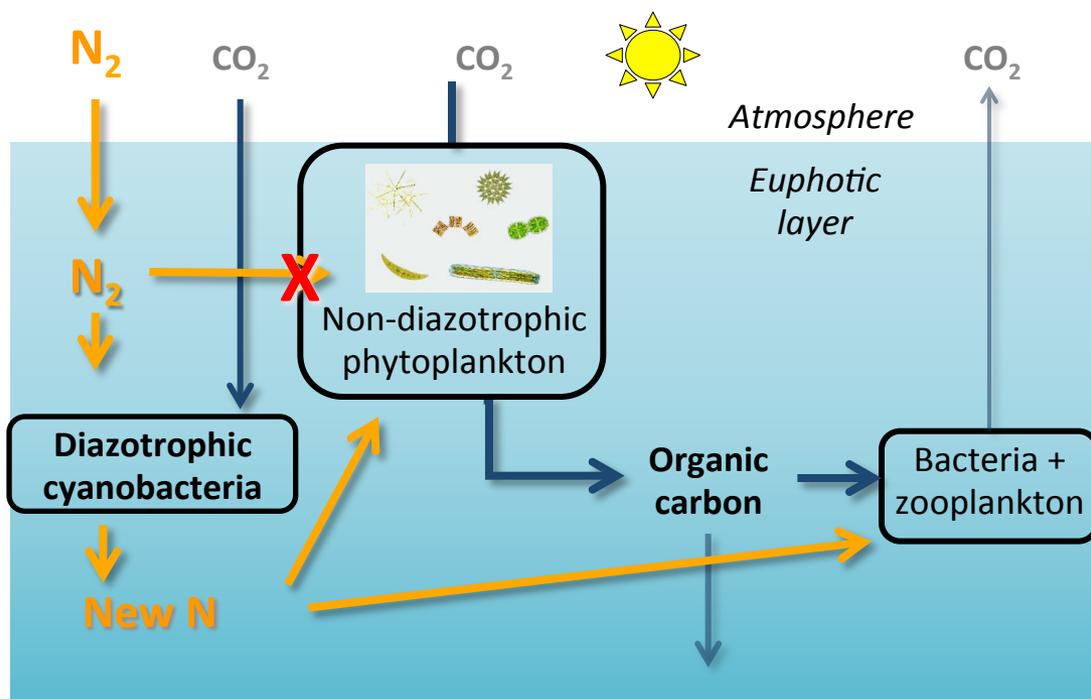
## N<sub>2</sub> fixation:

- ~50% of external N input
- ~50% of new production in some oligotrophic areas

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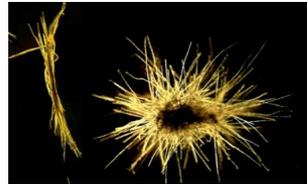
**Key role of N<sub>2</sub> fixation in biogeochemical cycles of C and N**

## **Diazotrophic cyanobacteria: key actors of oceanic N<sub>2</sub> fixation**

## Diazotrophic cyanobacteria: key actors of oceanic N<sub>2</sub> fixation

Filamentous

*Trichodesmium* spp.



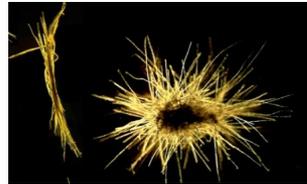
MIT/WHOI

**Has long been considered as dominant**

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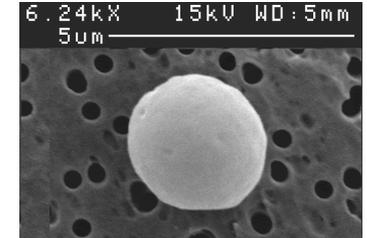


MIT/WHOI

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### Unicellulars (UCYN)

**UCYN-B**  
*Crocospaera watsonii*  
WH8501



(Jacq et al. 2014)

## Diazotrophic cyanobacteria: key actors of oceanic N<sub>2</sub> fixation

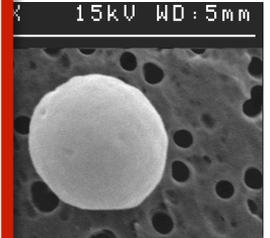
**UCYN:**

**~50% of global N<sub>2</sub> fixation** (*model, Monteiro et al. 2010*)

**Diazotrophes < 10µm:**

**≤ 75% of N<sub>2</sub> fixation in equatorial Pacific**

(*Bonnet et al. 2009*)



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# Diazotrophic cyanobacteria: key actors of oceanic N<sub>2</sub> fixation

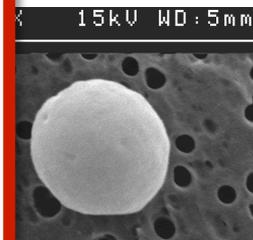
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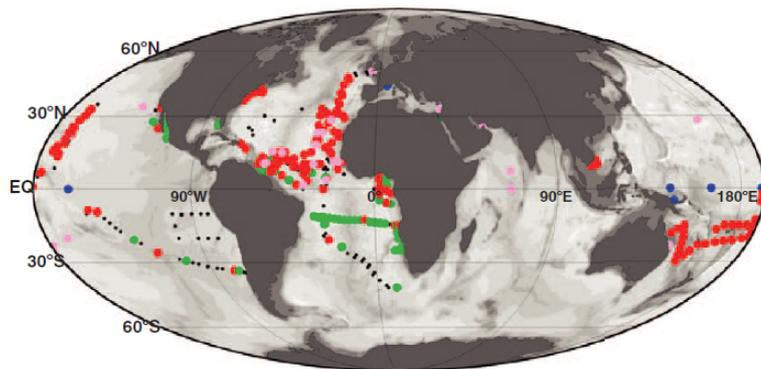


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Trichodesmium

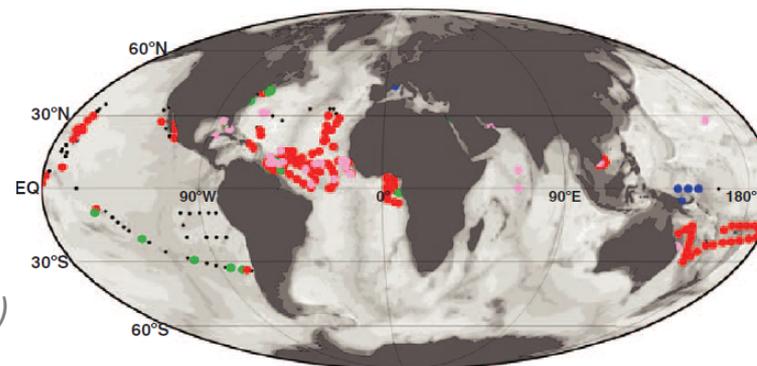
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### Trichodesmium spp.



(*Zehr 2011*)

### UCYN



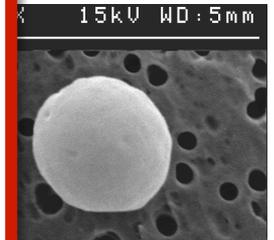
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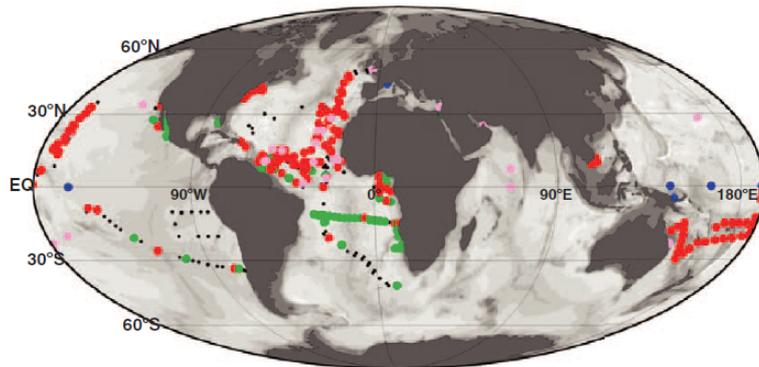


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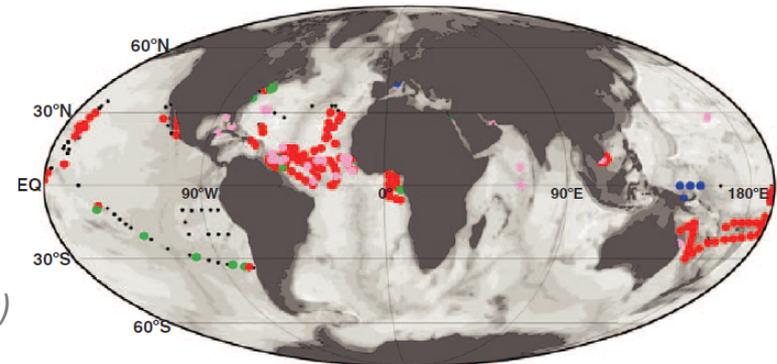
Has long been

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(*Zehr 2011*)

## UCYN



Coexistence of 2 types  
Mainly in (sub)tropical areas

## **Fe limitation of oceanic N<sub>2</sub> fixation**

- N<sub>2</sub> fixation = high bioenergetic cost → increase in photosynthesis
- Fe = cofactor of several photosynthetic enzymes

## Fe limitation of oceanic N<sub>2</sub> fixation

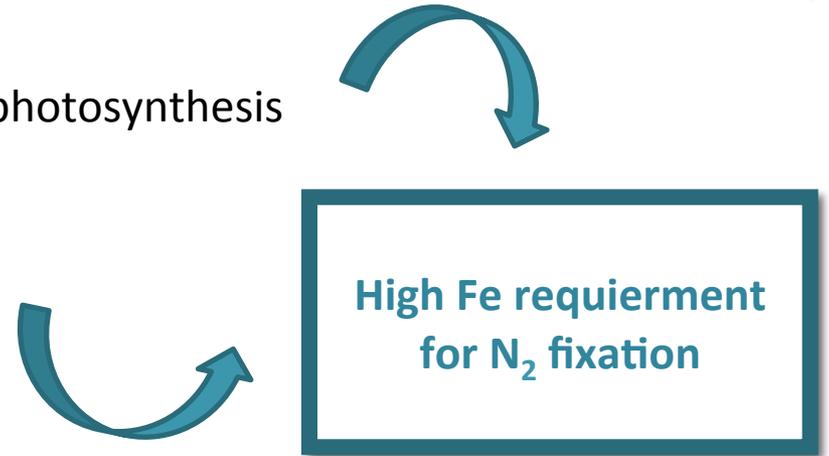
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**High Fe requirement  
for N<sub>2</sub> fixation**

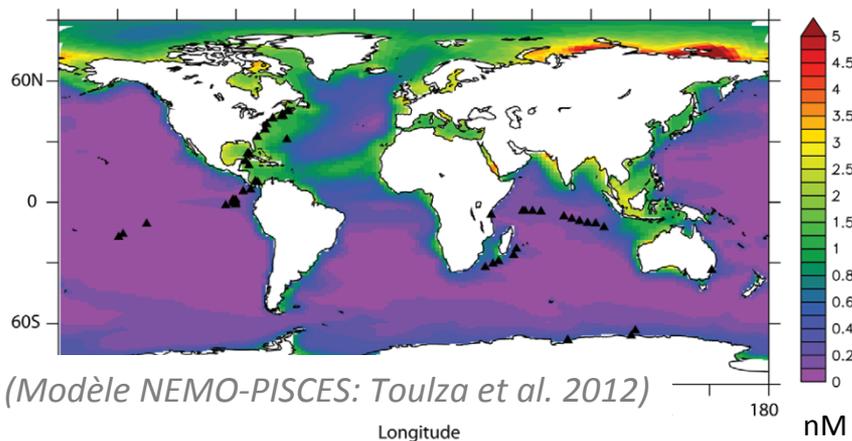
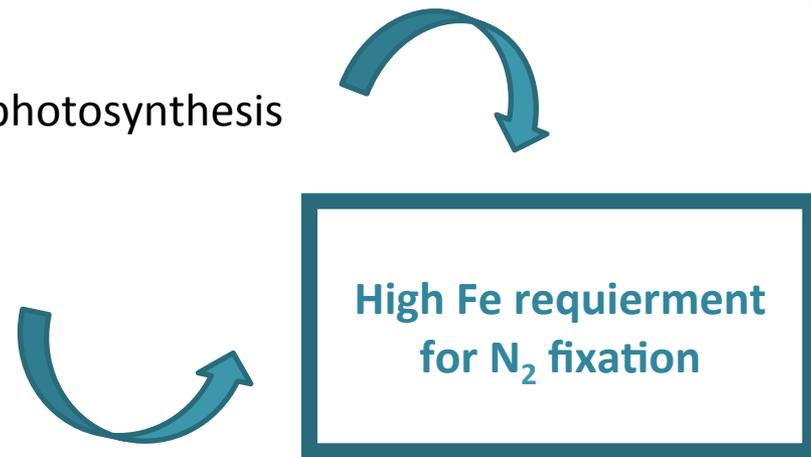
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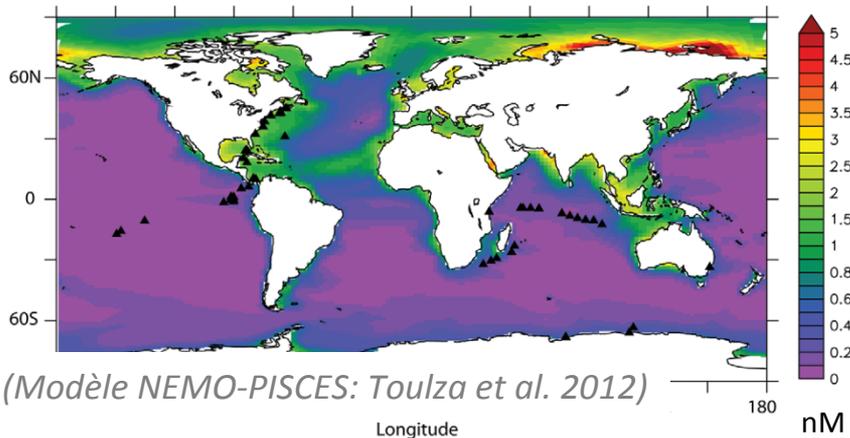
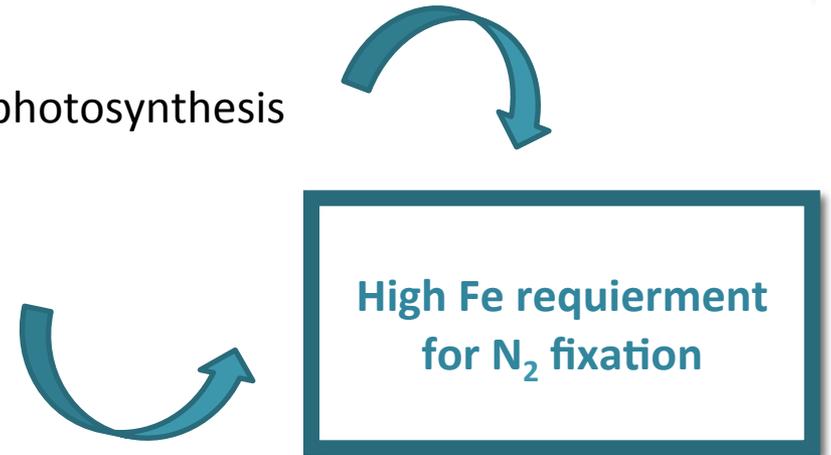


Very low [dFe] in surface open ocean  
( $< \sim 1$  nM)

→ **potential Fe limitation of N<sub>2</sub> fixation.**

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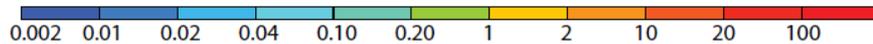
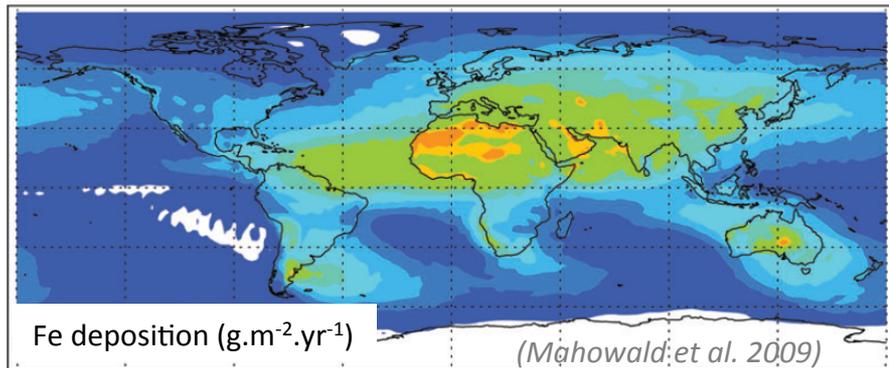
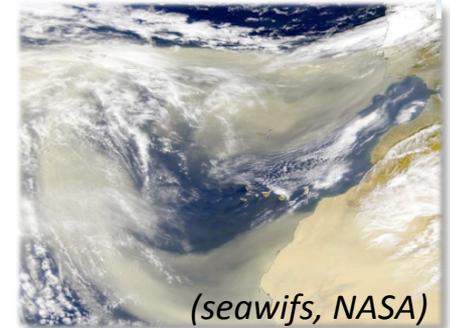
→ **potential Fe limitation of N<sub>2</sub> fixation.**

**Growth and activity of cultured *C. watsonii* are controlled by Fe concentration**

(Jacq et al. 2014)

## Atmospheric Fe deposition to the ocean

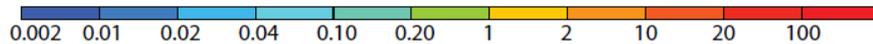
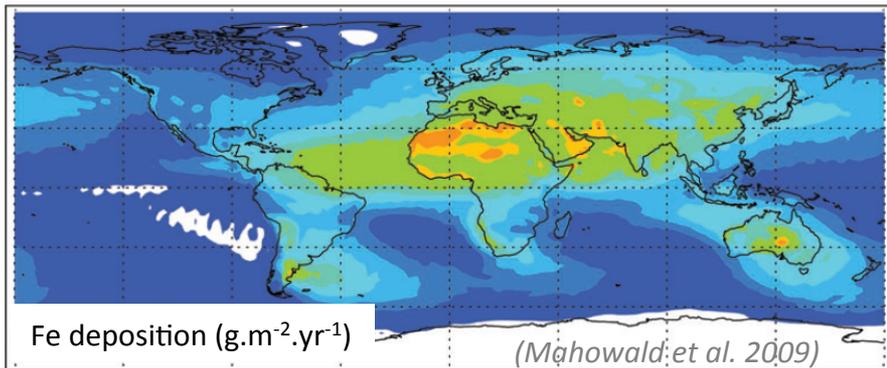
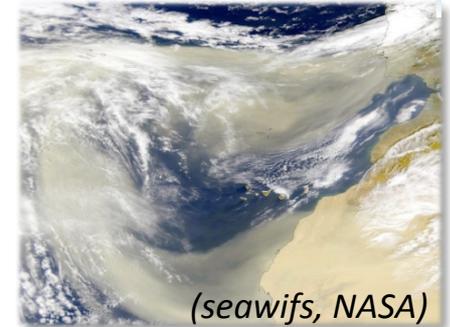
- Main Fe source to the surface open ocean
- Sahara: main atmospheric Fe source



## Atmospheric Fe deposition to the ocean

- Main Fe source to the surface open ocean
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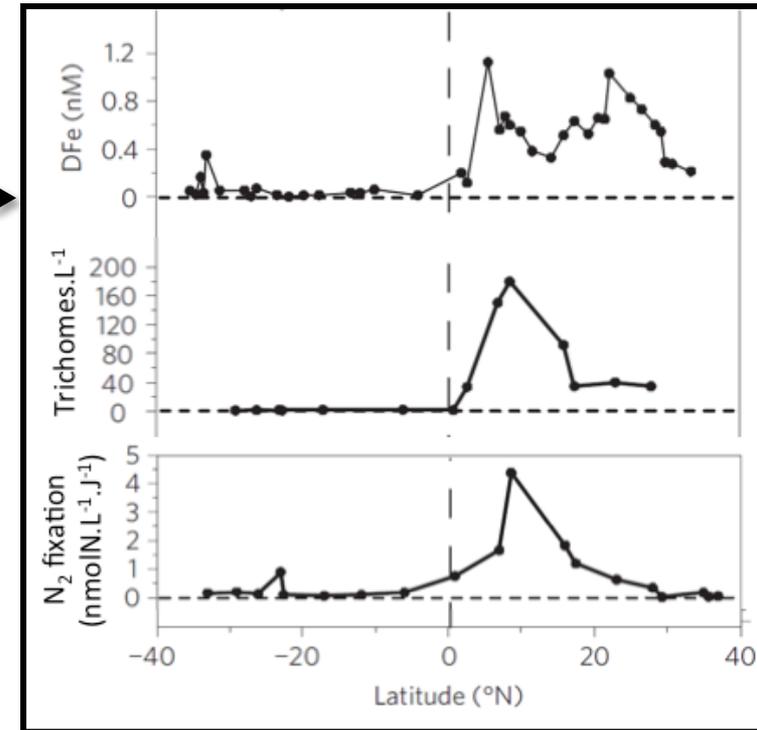
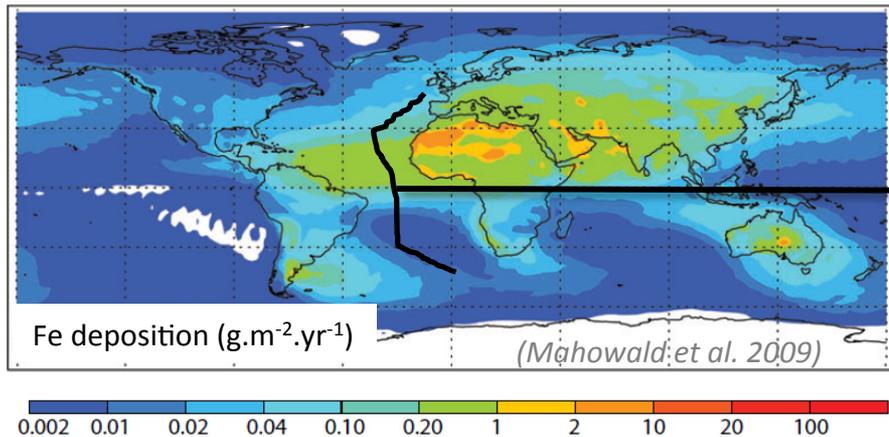
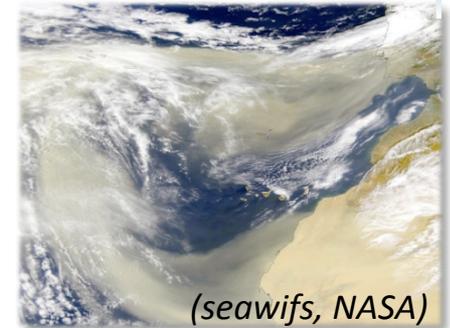
**Impact of atmospheric Fe on N<sub>2</sub> fixation still poorly known**



## Atmospheric Fe deposition to the ocean

- Main Fe source to the surface open ocean
- Sahara: main atmospheric Fe source

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(Moore et al. 2009)

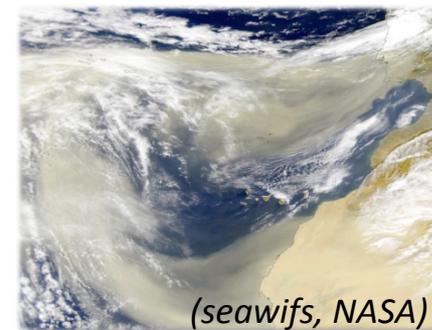
### Impact of atmospheric deposition on N<sub>2</sub> fixation:

Atlantic ocean:

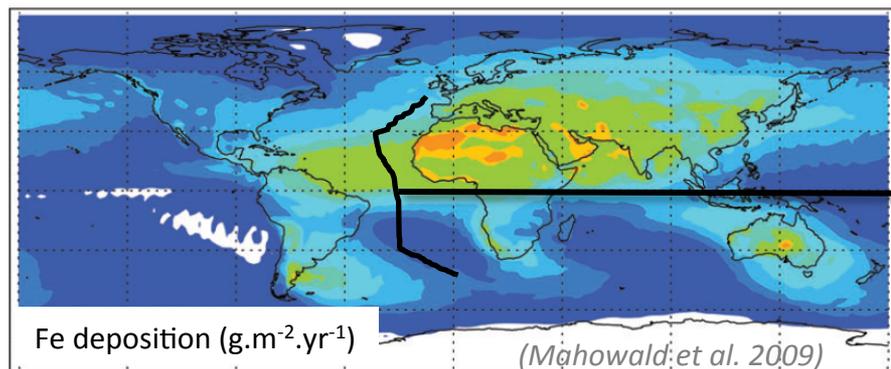
surface [dFe] concentrations, *Trichodesmium* abundance and N<sub>2</sub> fixation rates correlated

## Atmospheric Fe deposition to the ocean

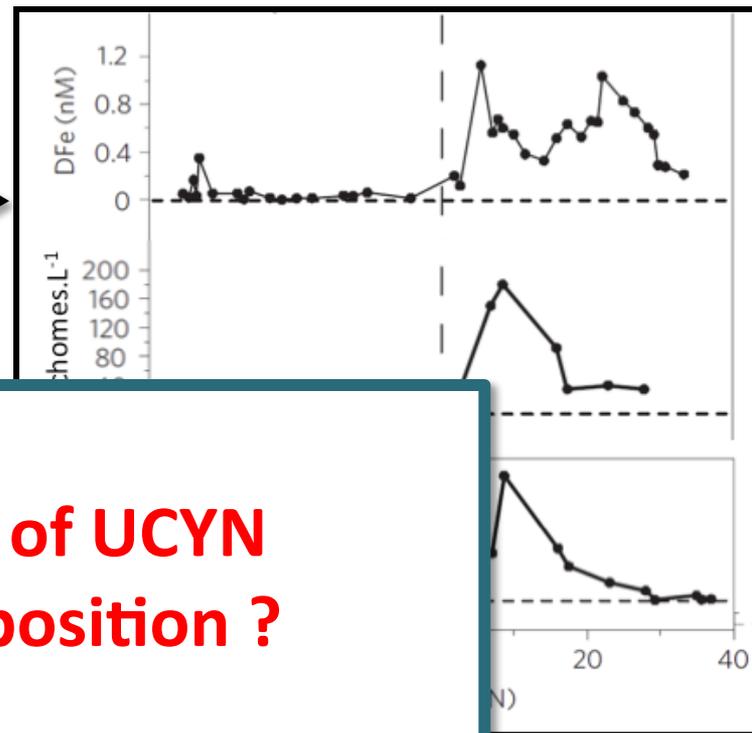
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Impact of atmospheric Fe on N<sub>2</sub> fixation still poorly known



0.002 0.01 0.0



(Moore et al. 2009)

What is the response of UCYN  
to atmospheric Fe deposition ?

Impact of atr

Atlantic oce

surface [dFe

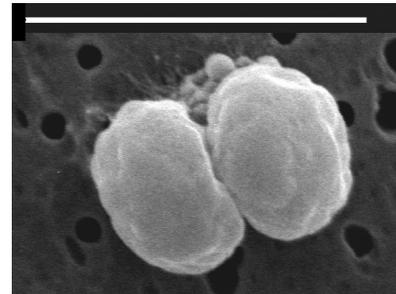
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## Experimental pan

Is Fe releasing by saharan dust abble to stimulate growth and activity of *C. watsonii* ?

Fe limited

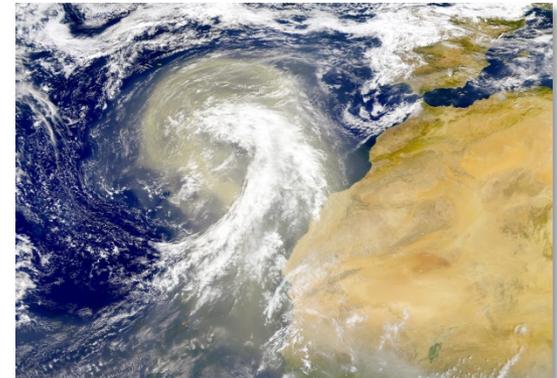
*C. watsonii* WH8501 cultures



*C. watsonii*

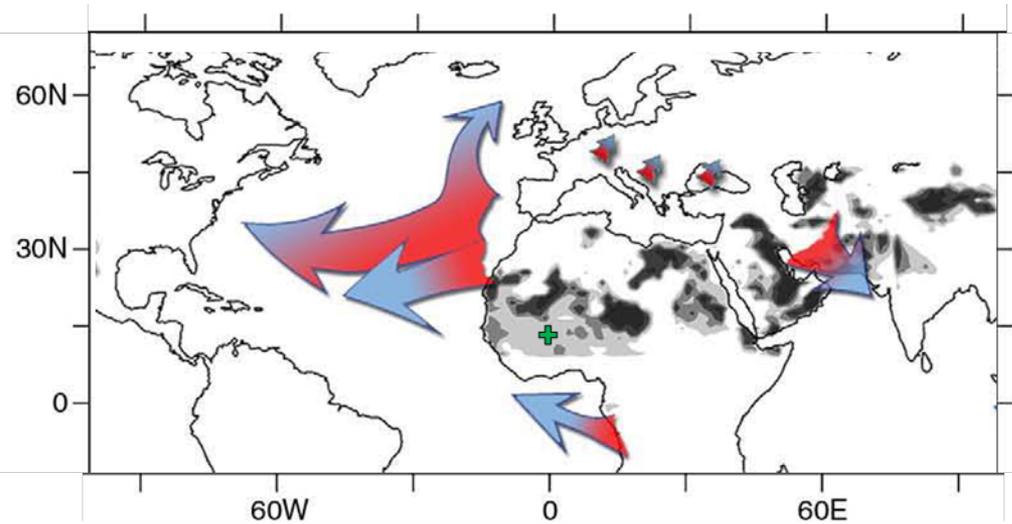


Addition of an artificial saharan rain



(seawifs, NASA)

## The Saharan dust

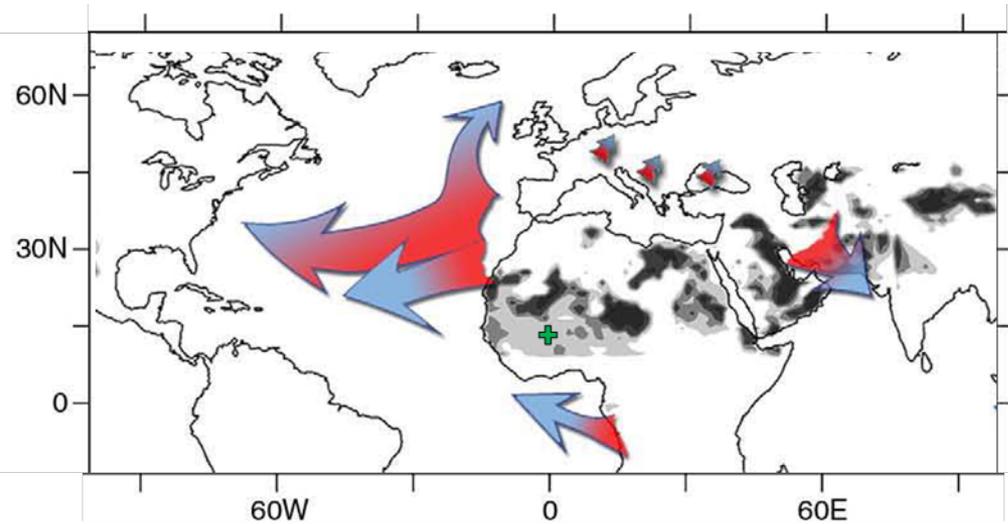


(Maher et al. 2010)

- Fine fraction ( $< 20 \mu\text{m}$ ) of Saharan soils (Niger)
- Représentative of collected aerosols in (sub)tropical North Atlantic
- Fe content:  **$3.1 \pm 0.2 \%$**

(Paris et al. 2011)

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## Artificial rain

Niger soils  $< 20 \mu\text{m}$

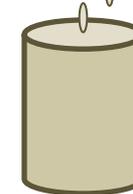


pH = 4.7  
**Oxalate:**  $1 \mu\text{M}$   
 Pc =  $250 \text{ mg.L}^{-1}$

1h – Natural light

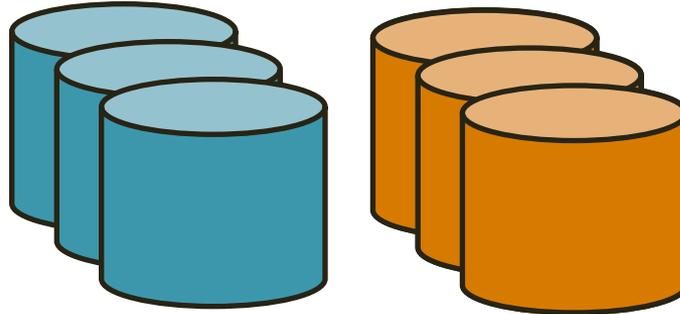


Filtration  $< 0.2 \mu\text{m}$



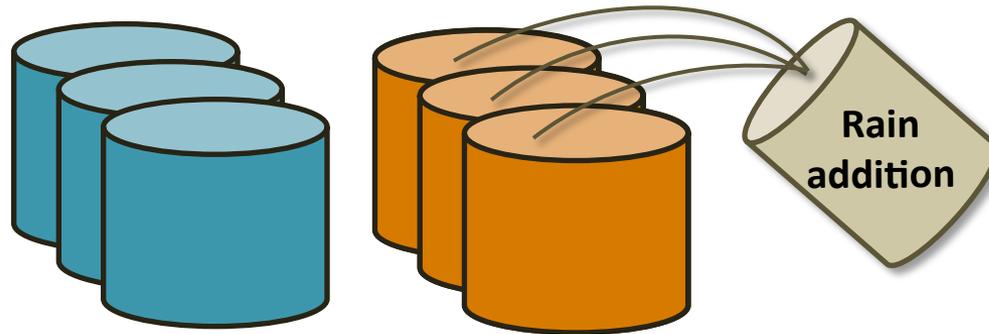
**dFe = 138 nM  $\rightarrow$  0.10 %**

## Experimental protocol



Control  
Ox = 200 nM  
dFe = 2nM

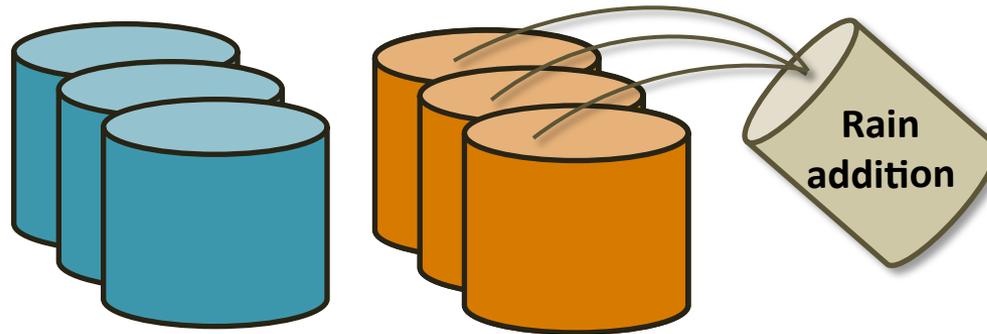
## Experimental protocol



Control  
Ox = 200 nM  
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**+ Rain**  
Ox = 200 nM + 20 nM  
dFe = 2 + **2.7 nM**

## Experimental protocol



Control

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dFe = 2 nM

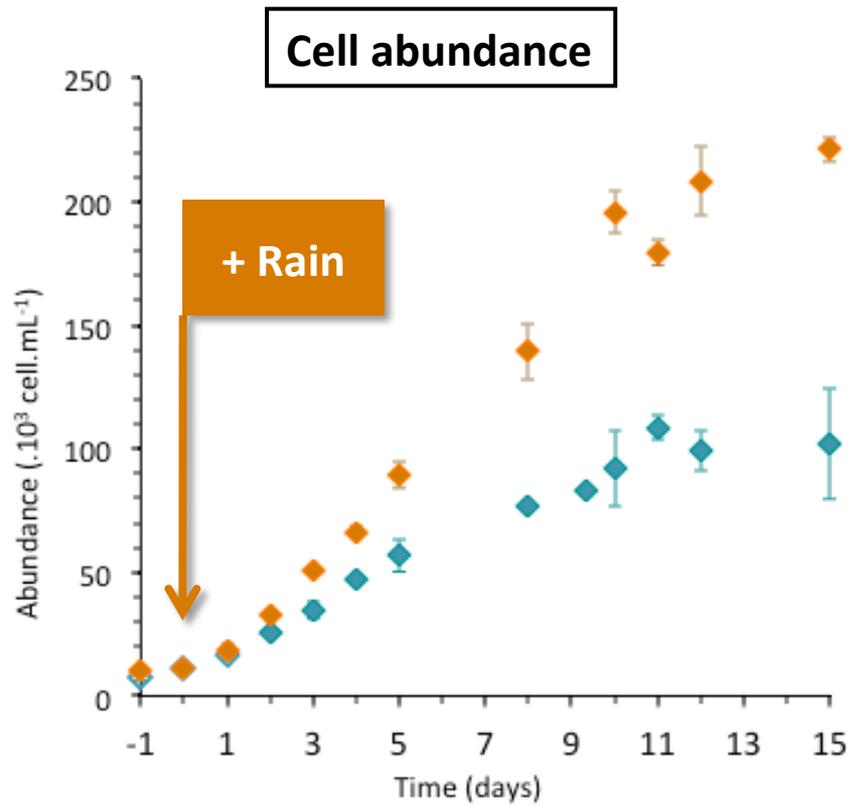
+ Rain

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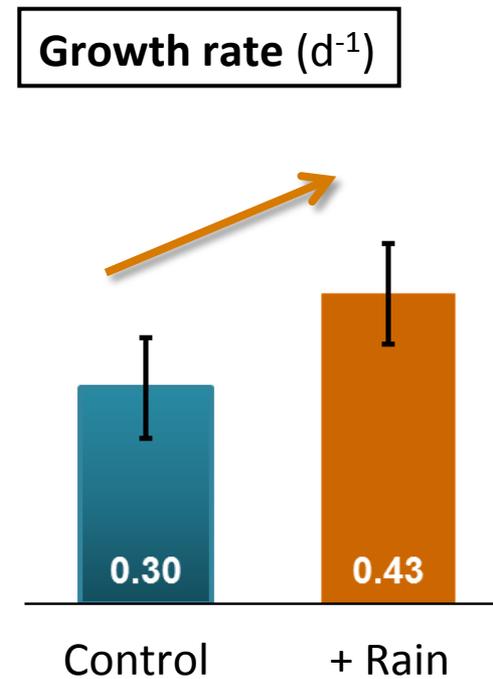
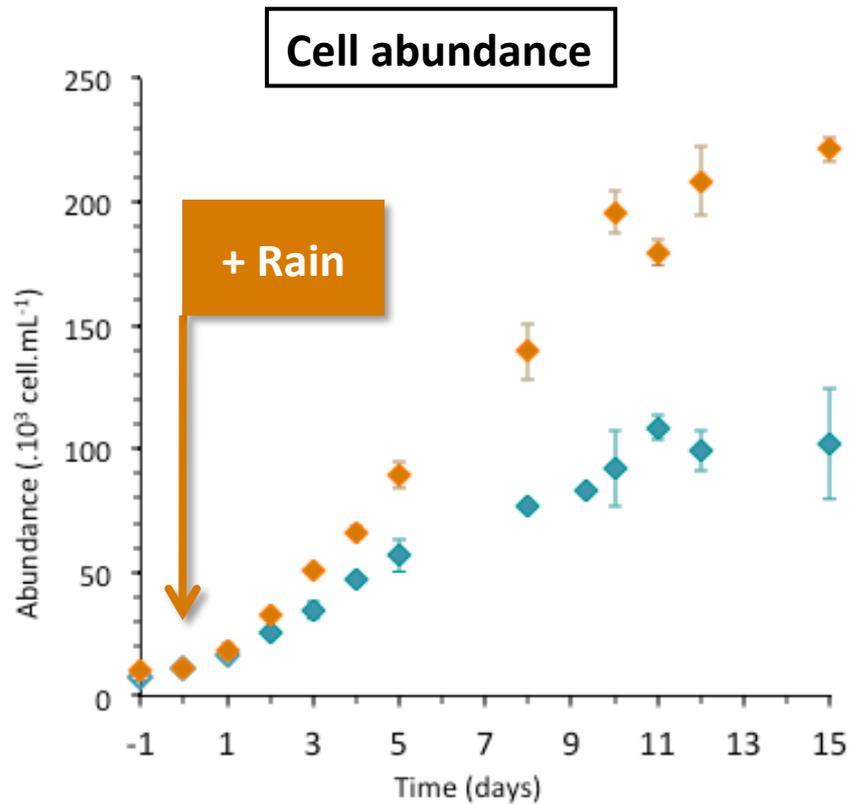
dFe = 2 + **2.7 nM**

- Cell abundance
  - C, N and Chl *a* cellular contents
  - N<sub>2</sub> fixation rates
  - CO<sub>2</sub> fixation rates
  - Cell volume
- } Triplicate measurement during exponential growth phase

## Physiological response

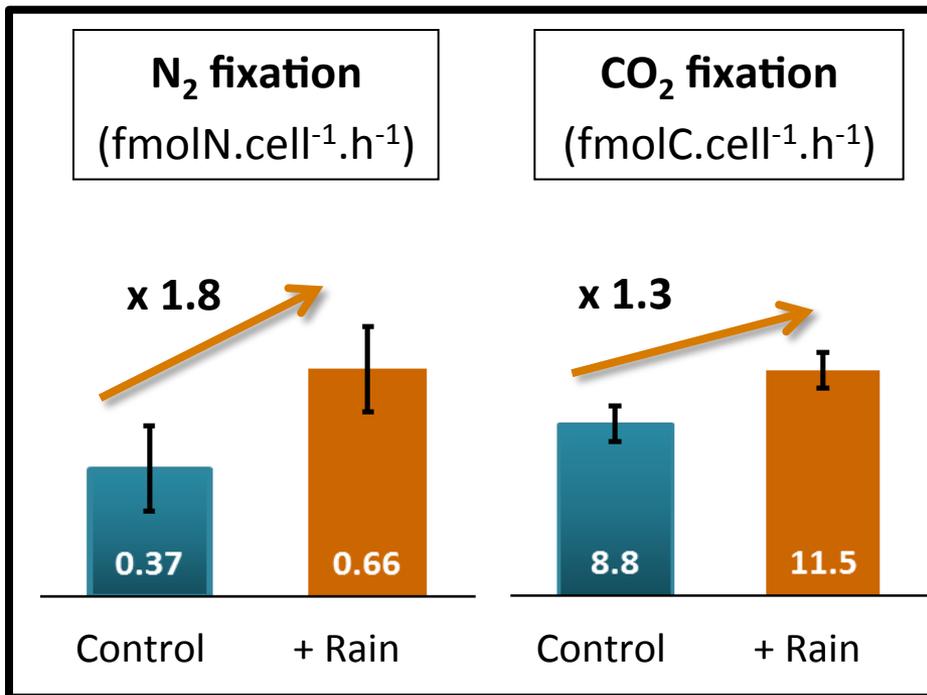


## Physiological response



**Dust released bioavailable Fe for *C. watsonii***

## Physiological response

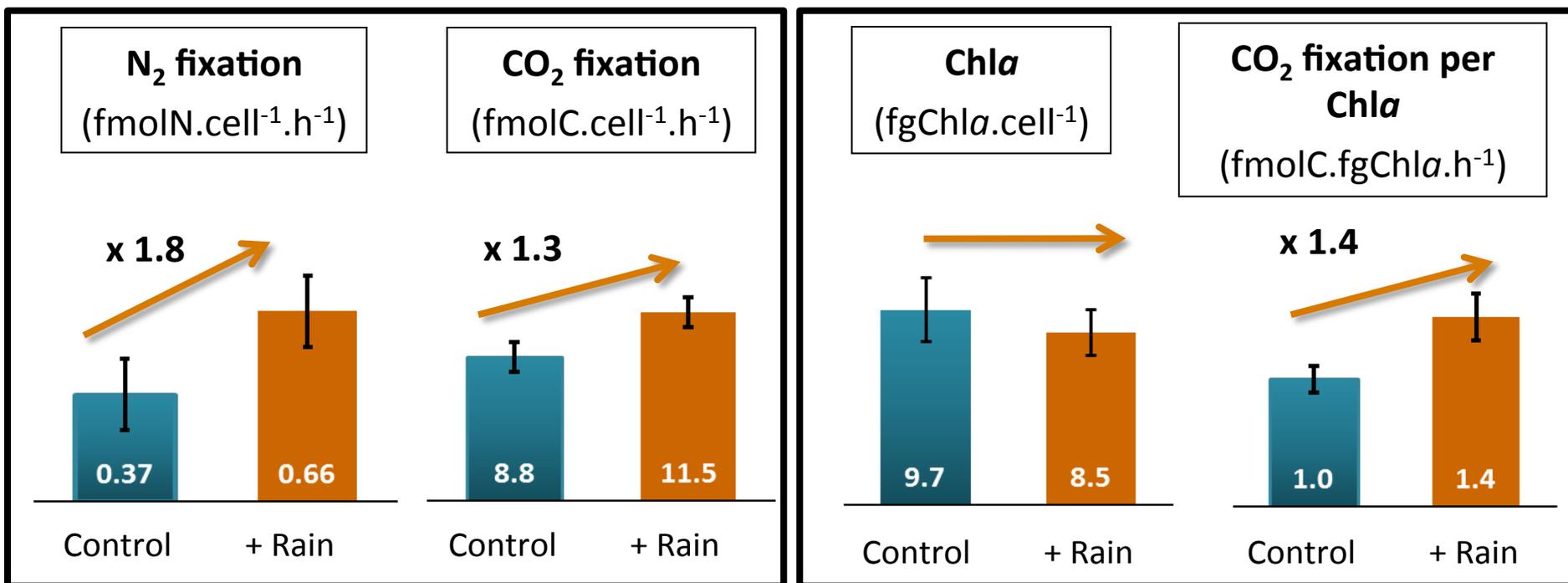


**Increase in synthesis of enzymes**

**Involved in N<sub>2</sub> fixation and photosynthesis**

**Higher impact on N<sub>2</sub> fixation  
than CO<sub>2</sub> fixation**

## Physiological response



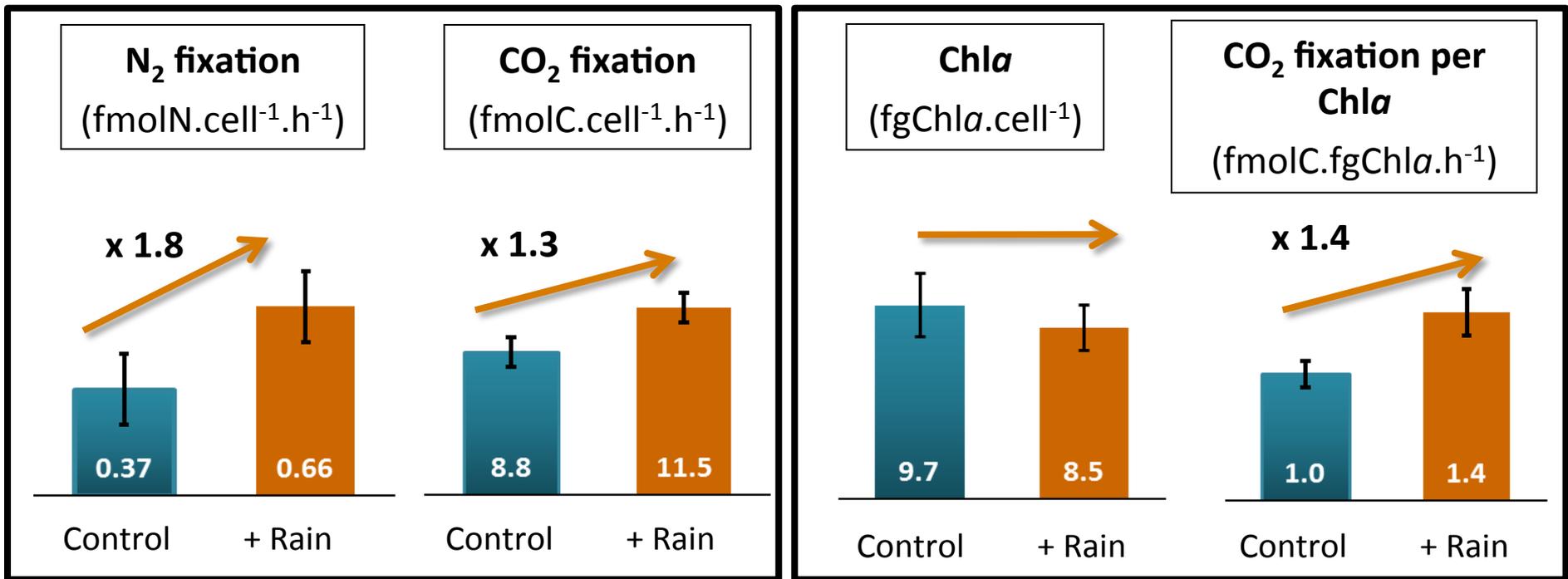
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Increase in  $CO_2$  fixation rate per Chl *a*

## Physiological response



Increase in synthesis of enzymes

Involved in  $N_2$  fixation and photosynthesis

Higher impact on  $N_2$  fixation  
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Increase in  $CO_2$  fixation rate per Chl a

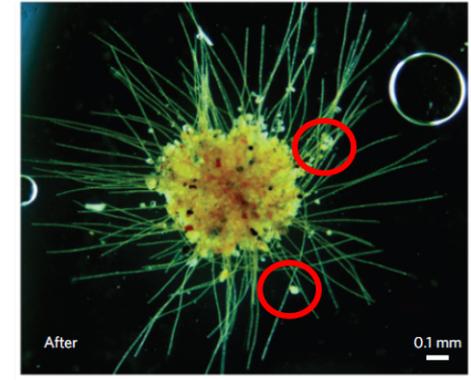
→ More efficient photosynthesis

↪ Increase in cytochromes synthesis?

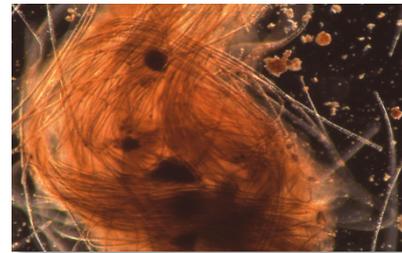
## *C. watsonii* versus *T. erythraeum*

Cultured *T. erythraeum*:

- *Rubin et al. (2011)*: Colonies → use directly particulate tanks to adsorption-dissolution mechanisms
- *Langlois et al. (2012)*: Formation of colonies following dust deposition and increase in cell abundance



*(Rubin et al. 2011)*

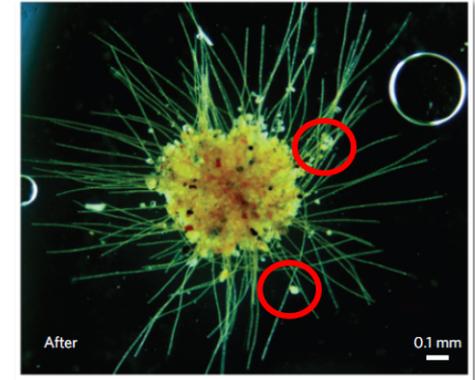


*(Langlois et al. 2012)*

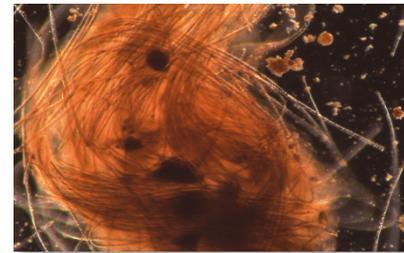
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(Rubin et al. 2011)



(Langlois et al. 2012)

**Positive response of 2 types of diazotrophic cyanobacteria  
→ Which would be favored in the ocean?**



*Thank you for your attention*

*and thanks to...*

*Céline Ridame  
and  
Alain Saliot*

*Karine Desboeufs*

*Fanny Kaczmar*

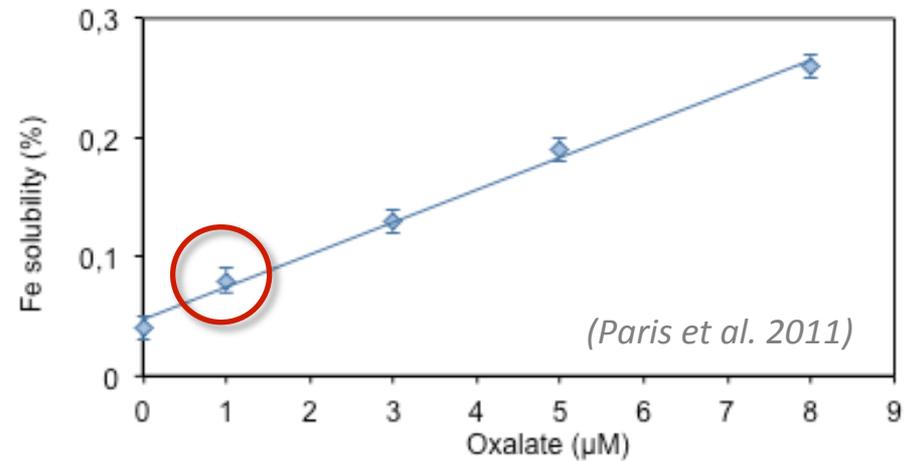
*Stéphane L'Heleguen*

*Hervé Rybarczyk*

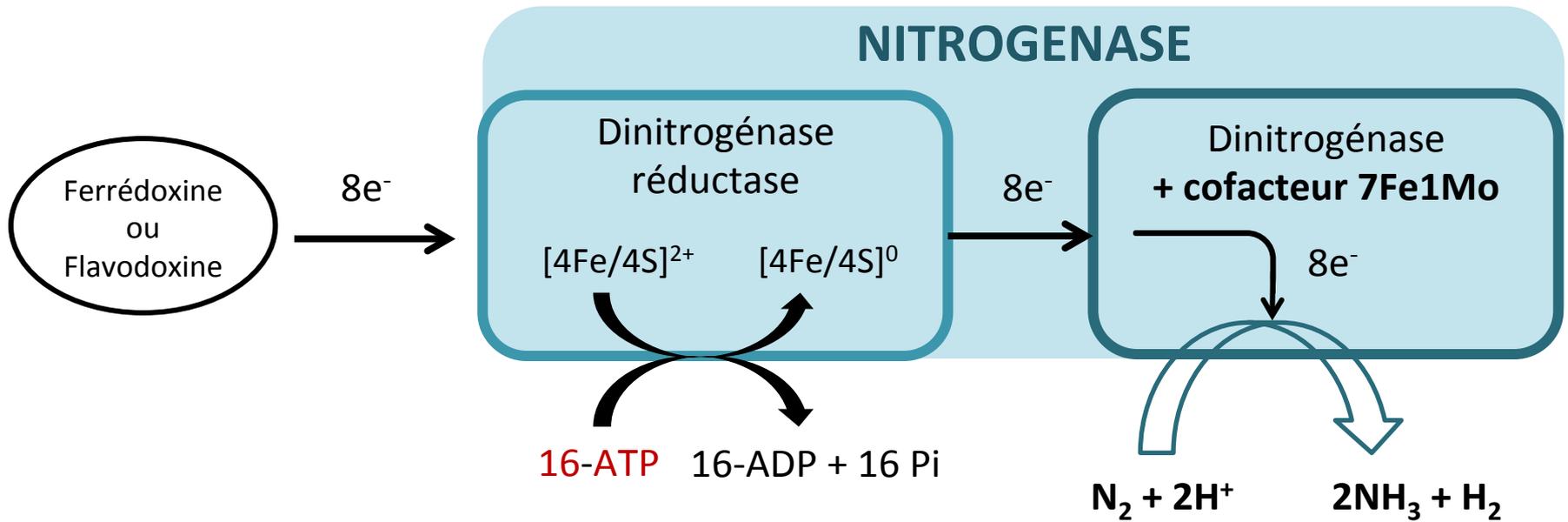
# Annexes

## Fe organic chelator

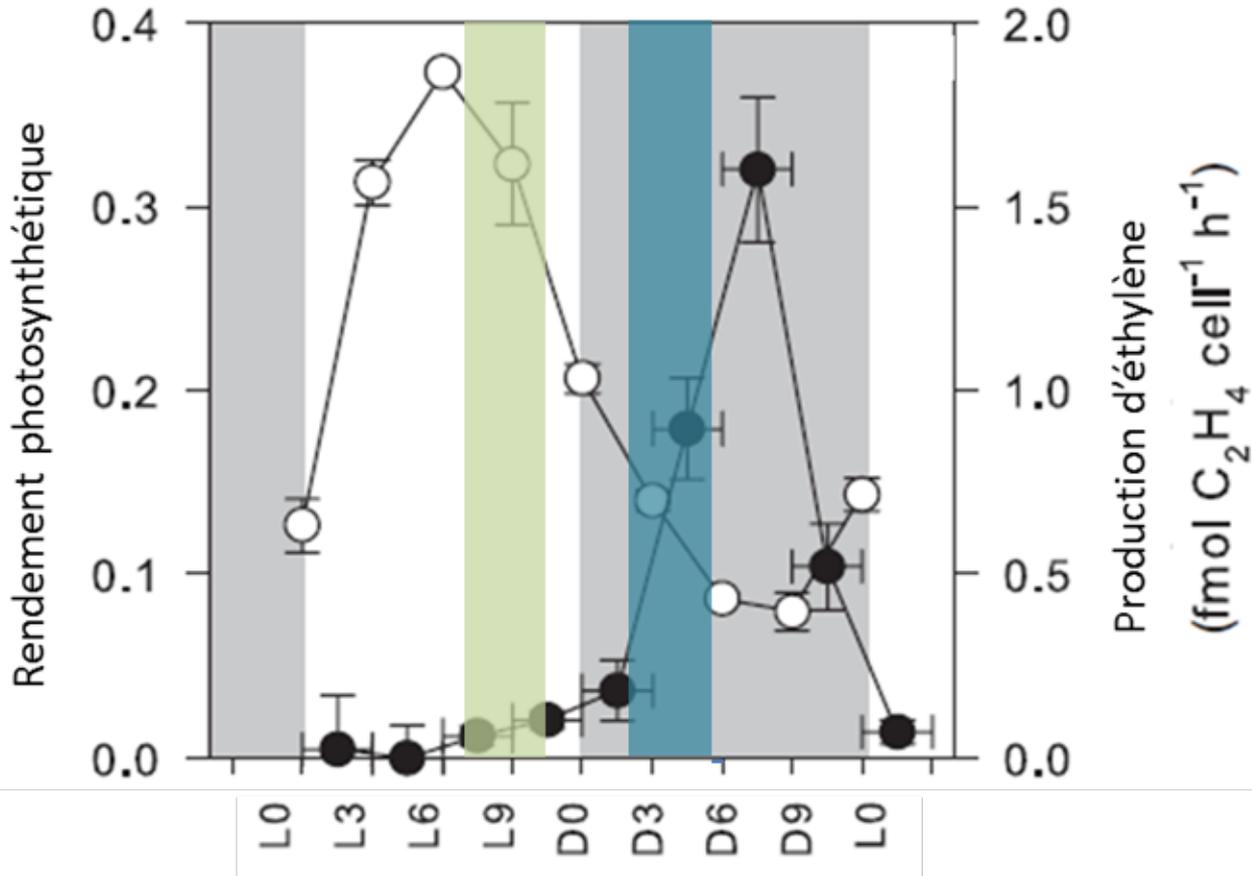
- **Oxalate**: the most abundant in rainwater and aerosols
  - Increase Fe solubility
  - Concentration in artificial rainwater: **1  $\mu\text{M}$**
- ➔ Theoretical Fe solubility: **0.08  $\pm$  0.01 %**



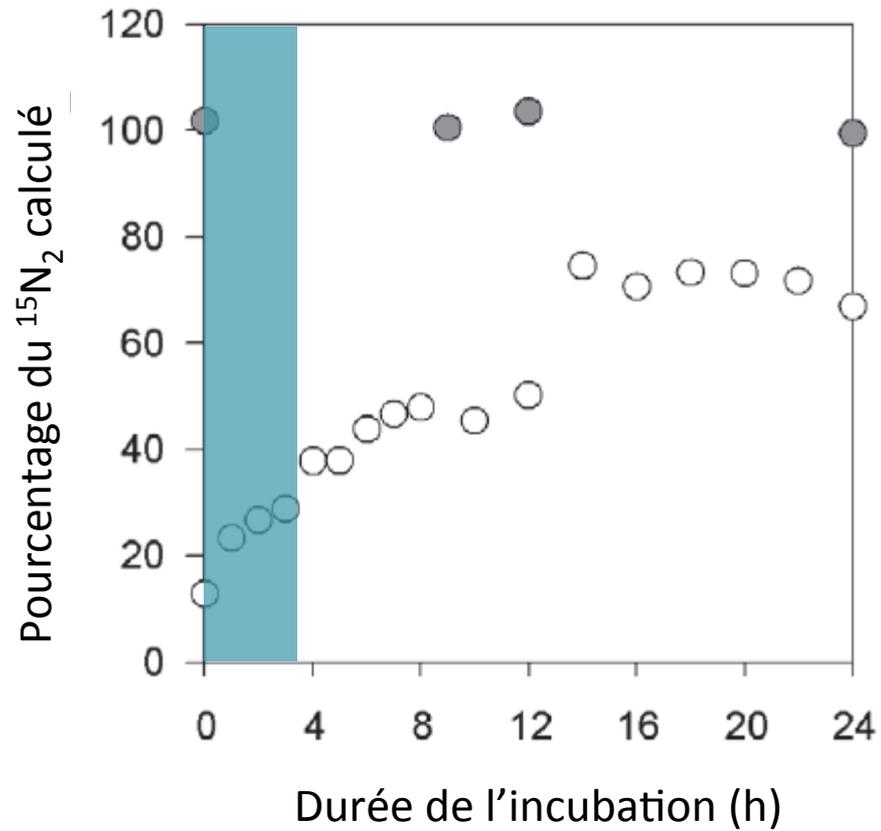
# La nitrogénase



# Prélèvements cultures: Fixation de CO<sub>2</sub> et N<sub>2</sub>



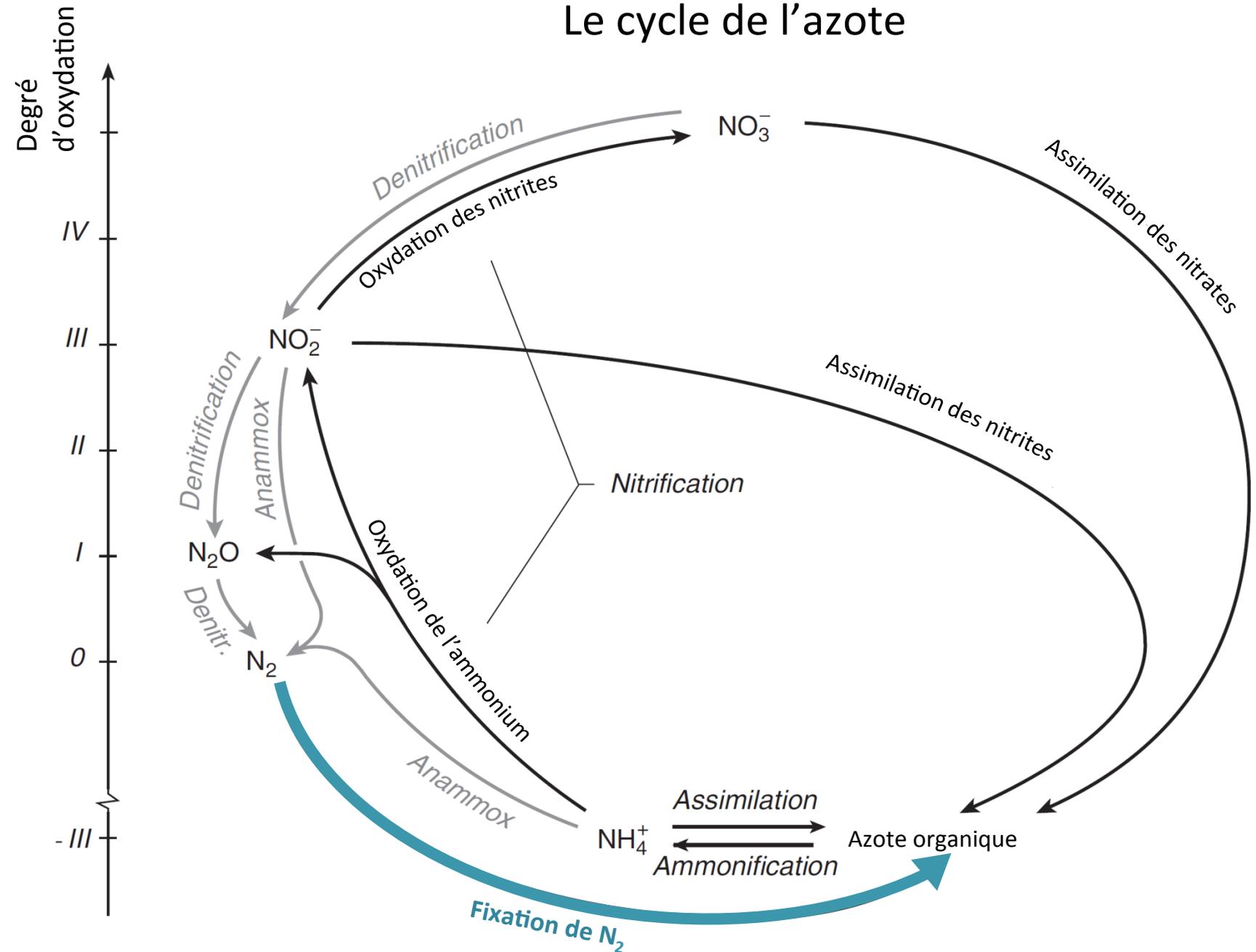
## Sous estimation des taux de fixation de N<sub>2</sub>



24%

Mohr et al. 2010b

# Le cycle de l'azote



# Chélation du fer

**pFe > dFe car peu soluble dans l' eau de mer**



**99% du dFe sous forme  
complexée**

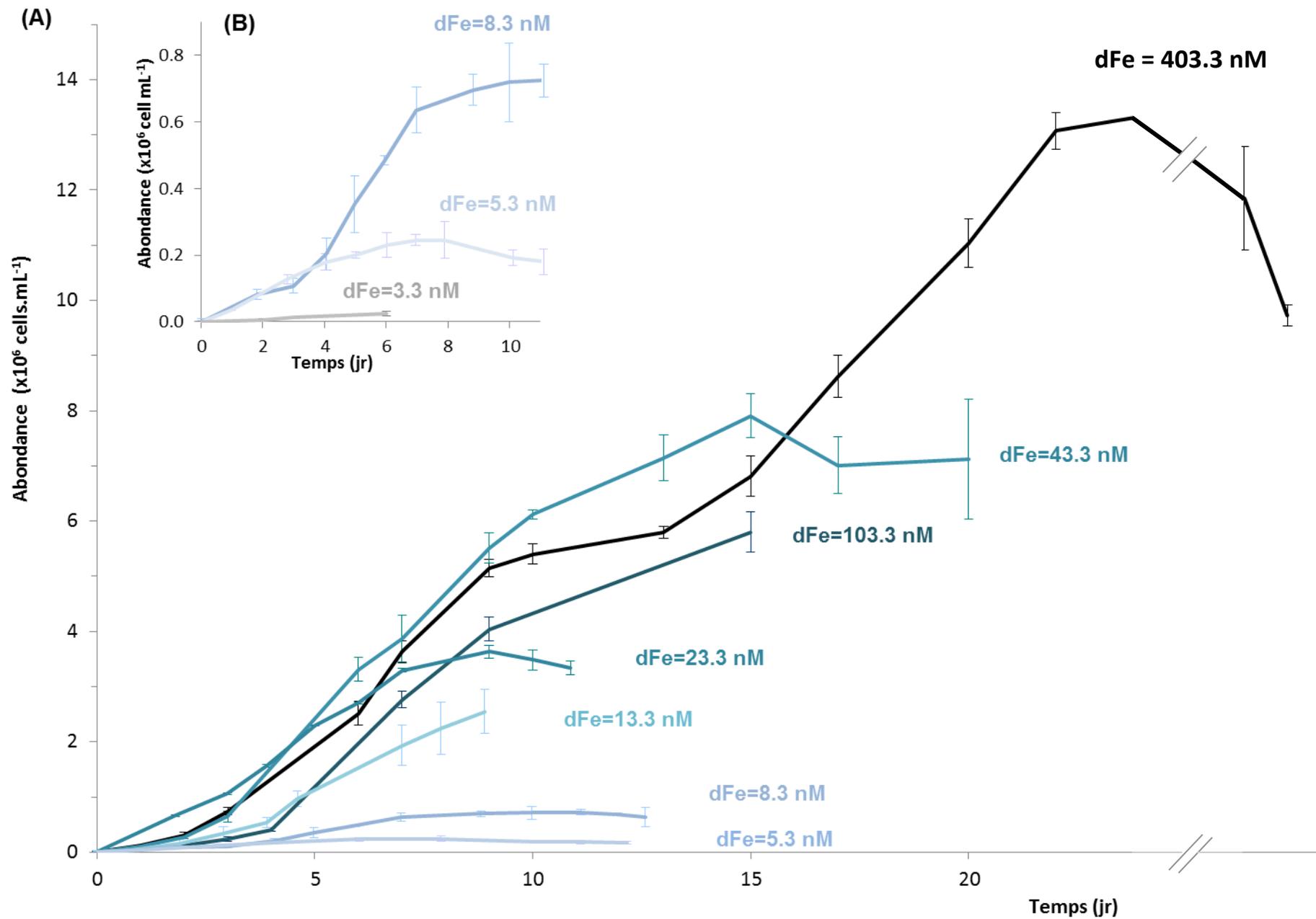
(Gledhill et Van Den Berg 1994)

**Fer organique dissous**

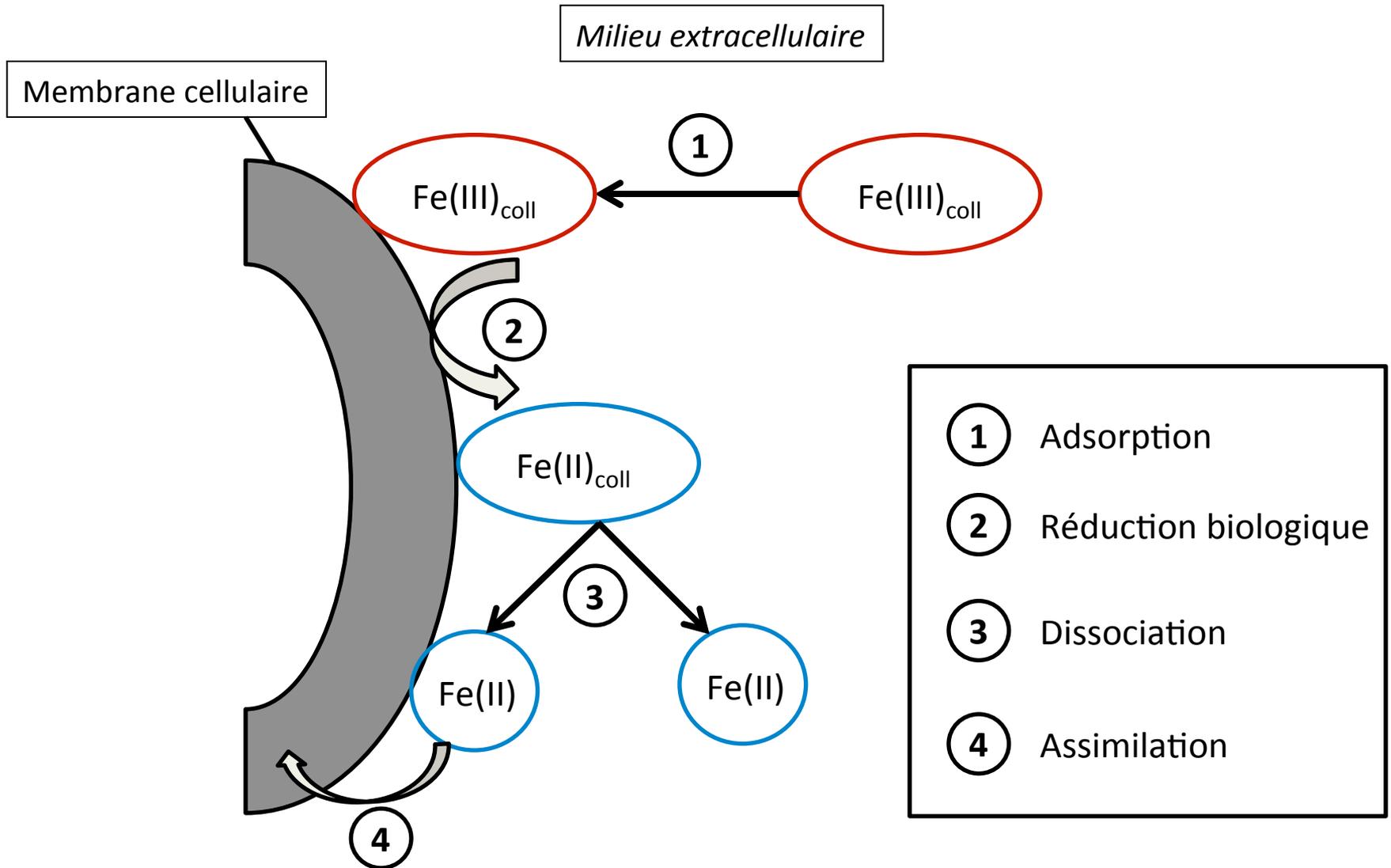


Très faible solubilité

(Millero, 1998)



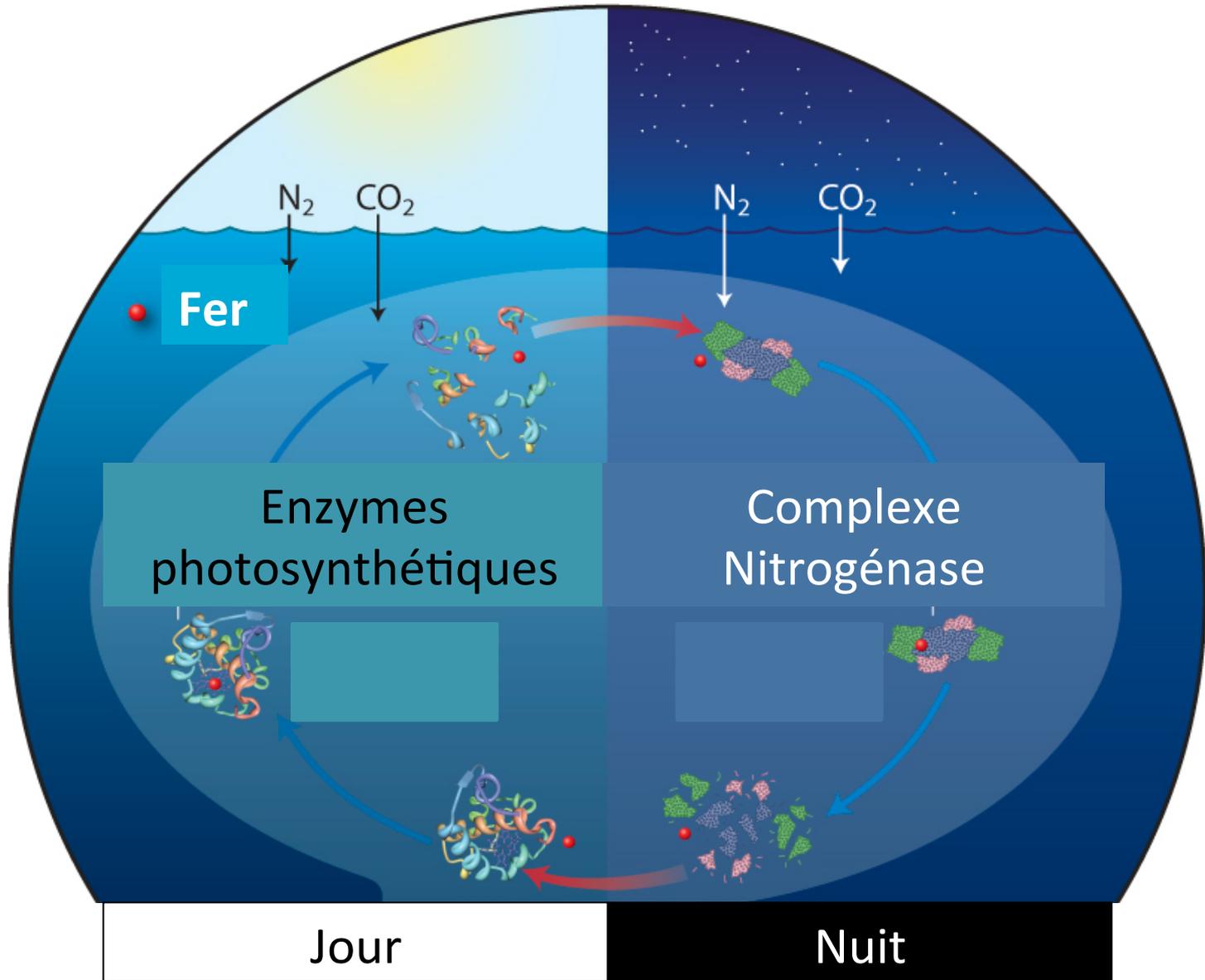
# Mécanisme d'assimilation du fer colloïdal



Sunda, comm. pers.

# Stratégie de recyclage du fer

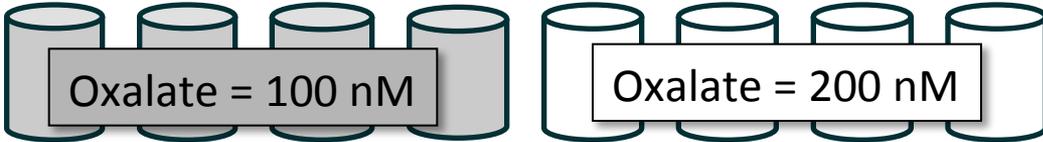
Saito et al. 2011



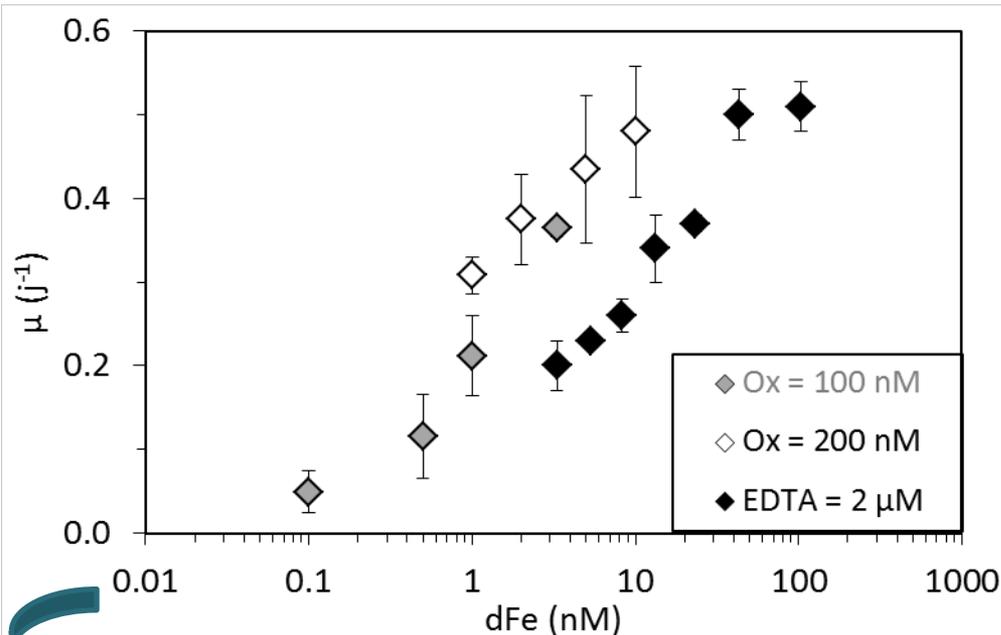
## 2. Réponse de *C. watsonii* face à un apport humide saharien

### L'impact de l'oxalate ( $C_2O_4^{2-}$ ) comme agent chélateur du fer sur la croissance de *C. watsonii*

*C. Watsonii*  
Milieu YBCII modifié  
Conditions  
« trace metal clean »



	Oxalate = 100 nM				Oxalate = 200 nM			
dFe (nM)	0.1	0.5	1	3.3	1	2	5	10
Fe' (nM)	0.04	0.11	0.26	0.80	0.52	0.68	0.63	1.50



- Oxalate vs. EDTA:  
Taux de croissance plus élevés
- Oxalate 100 nM vs. Oxalate 200 nM:  
Taux de croissance plus élevé



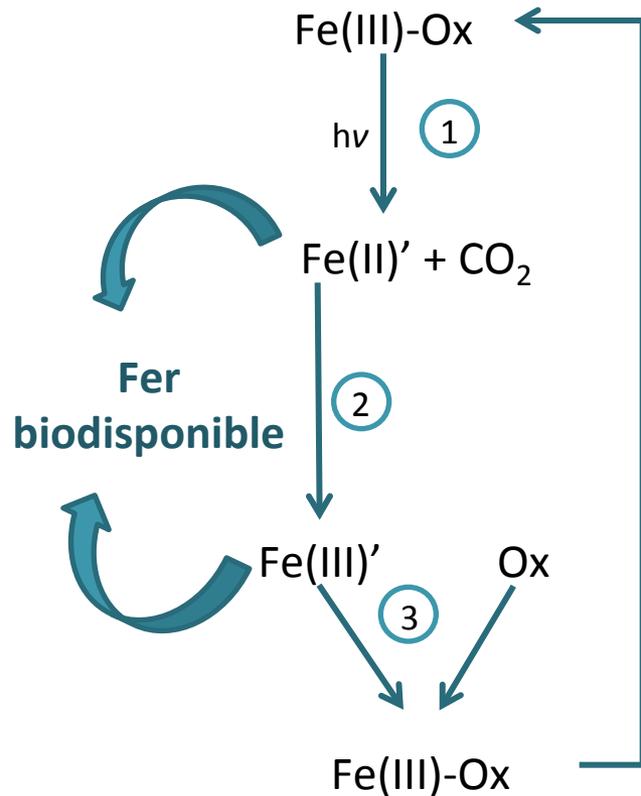
- Utilisation de l'équation de Monod pour déterminer Fe' :

$$[Fe'] = K_{\mu Fe'} \times \frac{\mu}{\mu_{\max} - \mu}$$

$$K_{\mu Fe'} = 0.38 \pm 0.11 \text{ nM}$$

Augmentation des concentrations en Fe' ?

### L'oxalate comme agent chélateur du fer dans l'eau de mer: (Bill Sunda, com. pers.)



Dans l'eau douce:

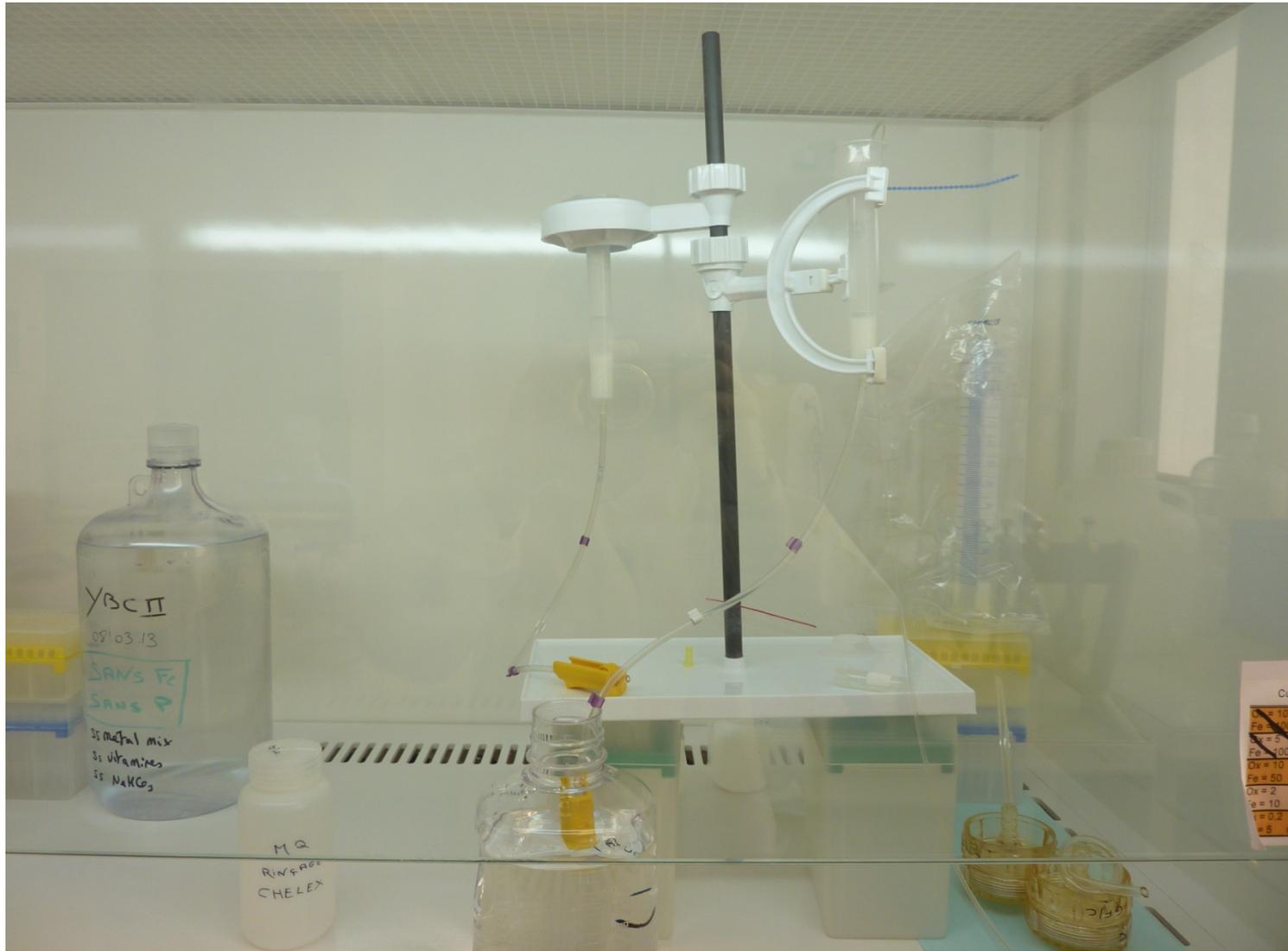
constantes de complexation Fe-oxalate < Fe-EDTA

→ De même dans l'eau de mer?

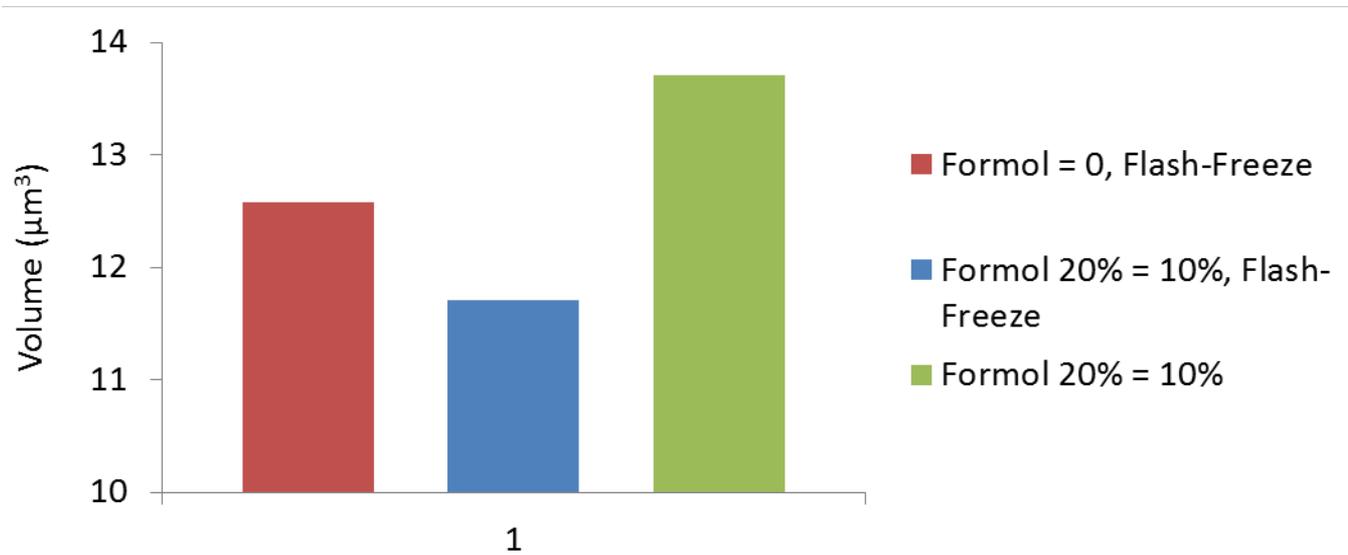
- 1 Oxydo-réduction photochimique
- 2 Oxydation
- 3 Complexation

- Biodisponibilité, au moins partielle, du complexe fer-oxalate?

# Montage CHELEX LOCEAN



# Influence du fixateur



## Biogeochemical relevance

⊕ Increase in desert areas  
⊕ Precipitations ?

Wet saharan  
deposition

⊕ Intensification of stratification

Atmosphere

Euphotic  
layer

