

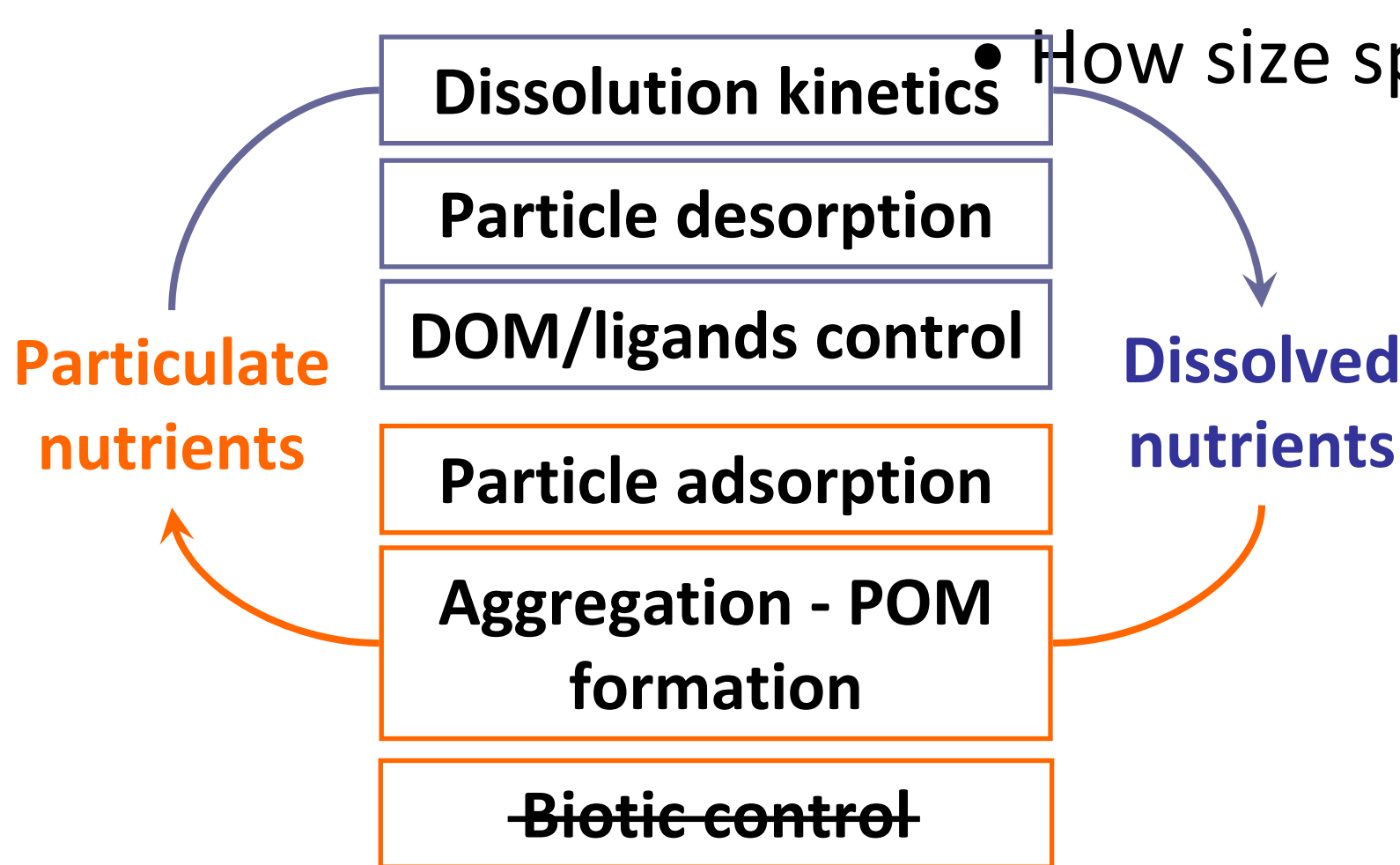
Impact of Saharan dust deposition on dissolved-colloidal-particulate nutrient distribution in seawater

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Objectives

- Quantify dissolution and adsorption processes in **abiotic condition** with settling of particles representative of dynamic in natural system



- Dissolved organic matter (DOM)** = positive feedback between dust and ocean biogeochemistry?
- Seasonal variation:** net effect of atmospheric deposition dependent of seawater characteristics?

Methods



- 300 L of seawater collected and filtered (0.2 μm) at 3 periods: **bloom – post-bloom – winter mixing**
MAY **OCTOBER** **FEBRUARY**
- Seeding: wet deposition (10 g.m⁻²) of processed Saharan dust (Guieu et al. 2010)
- Sampling and handling under ultra-clean conditions
- Discrete measurements at 3 depths (0.1 – 0.3 – 0.6 m)**
PSD = 0.7 – 250 μm (Coulter Counter + LISST-100)
P – Fe : analysis after filtration at 0.2 and 0.015 μm
 a_{CDOM} – TEP – (DOC)

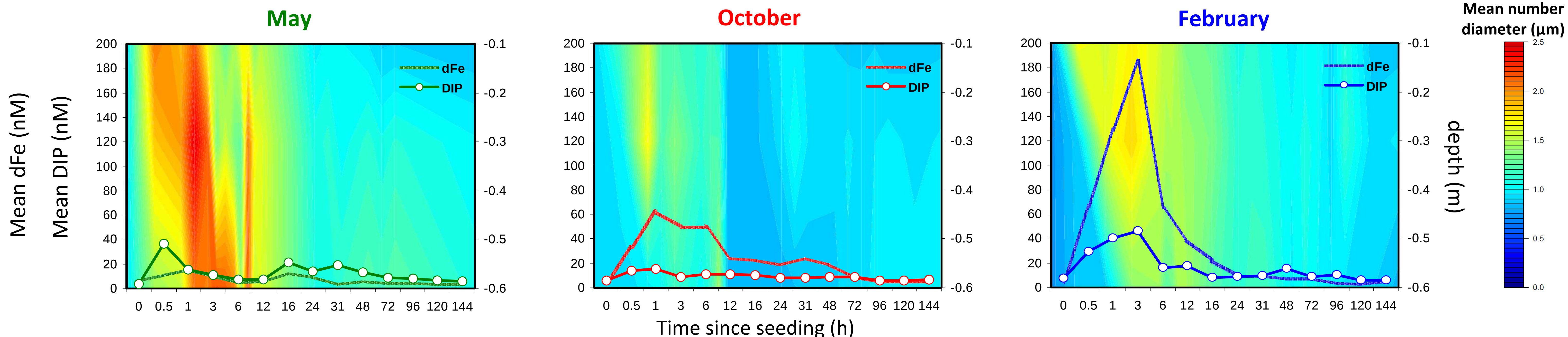
Contrasted initial DOM conditions

	MAY	OCTOBER	FEBRUARY
TEP [$\mu\text{mol/L}$]	27.1	16.3	2.7
$a_{\text{CDOM}}(300)$ [m^{-1}]	0.403	0.357	0.312
dFe [nM]	3.7	3.7	3.9
DIP [nM]	3.5	6	9

TEP and a_{CDOM} values reveal different initial quantities and qualities of DOM: **May** > **October** > **February**

What is the role of the DOM in the dissolution of atmospheric Fe and P after a Saharan dust event?
(colloids – aggregation process?)

Seasonal Variability of atmospheric Fe and P Dissolution: RESULTS



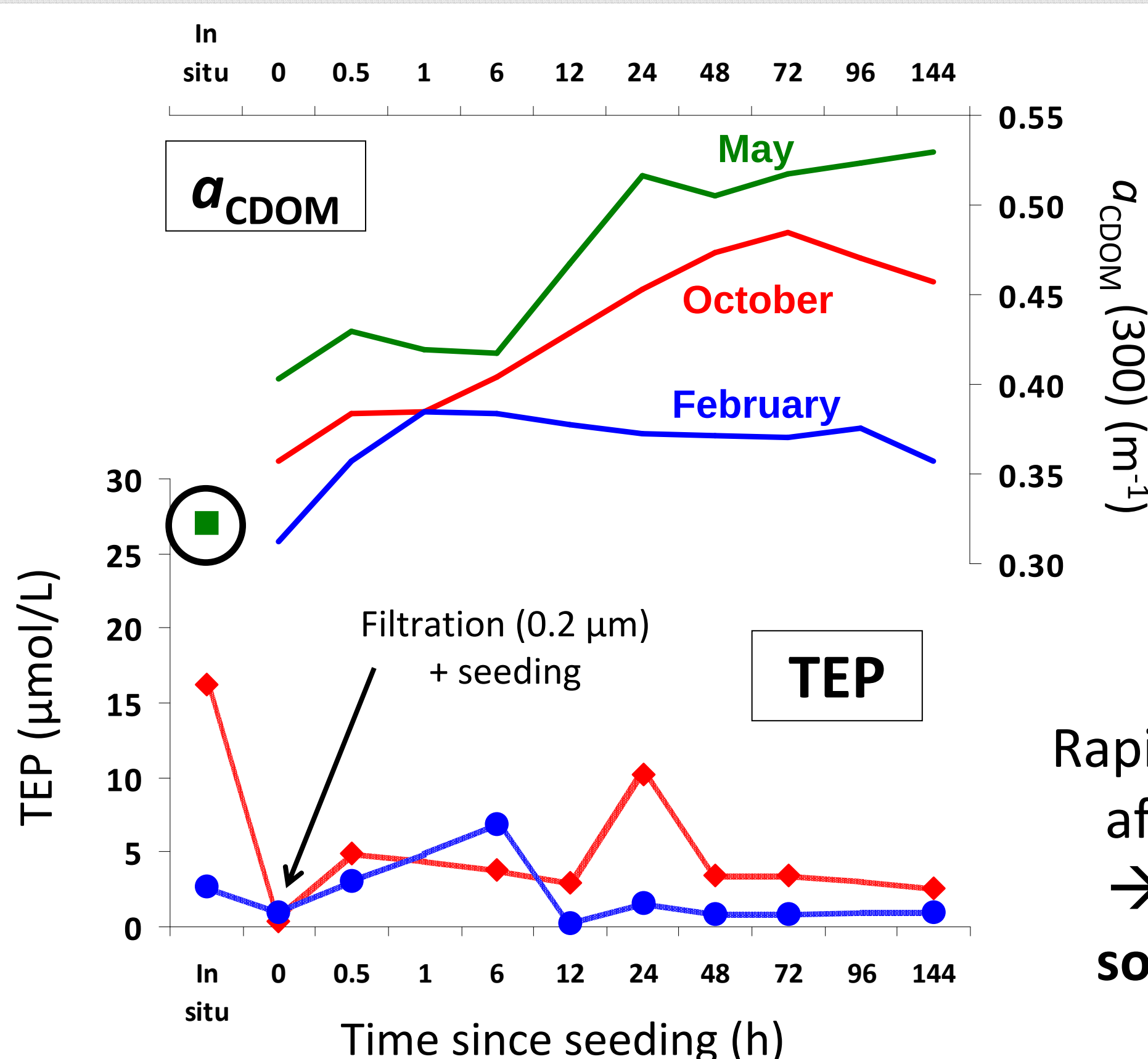
Seasonal variation of DOM → Variation in the dissolution of Fe and P (“easily” dissolvable Fe and P fraction) - after one week: same situation

Evidence of aggregation → in May and October, number of collisions with colloids (DOM) was higher leading to significantly lower net dissolution

Very fast kinetic processes → few hours (6 – 12 h) after seeding, the decrease of [dFe] and [DIP] corresponds to scavenging (no biological uptake) (Wagener et al. 2010)

Saharan dust event when high *in situ* DOM → lower supply of new dissolved nutrients to the system

What control the fate of atmospheric Fe and P?



May and October: Abiotic formation of colloids and aggregates after seeding

TEP formed from filtered precursors (< 0.2 μm)
(data not available in May after seeding)

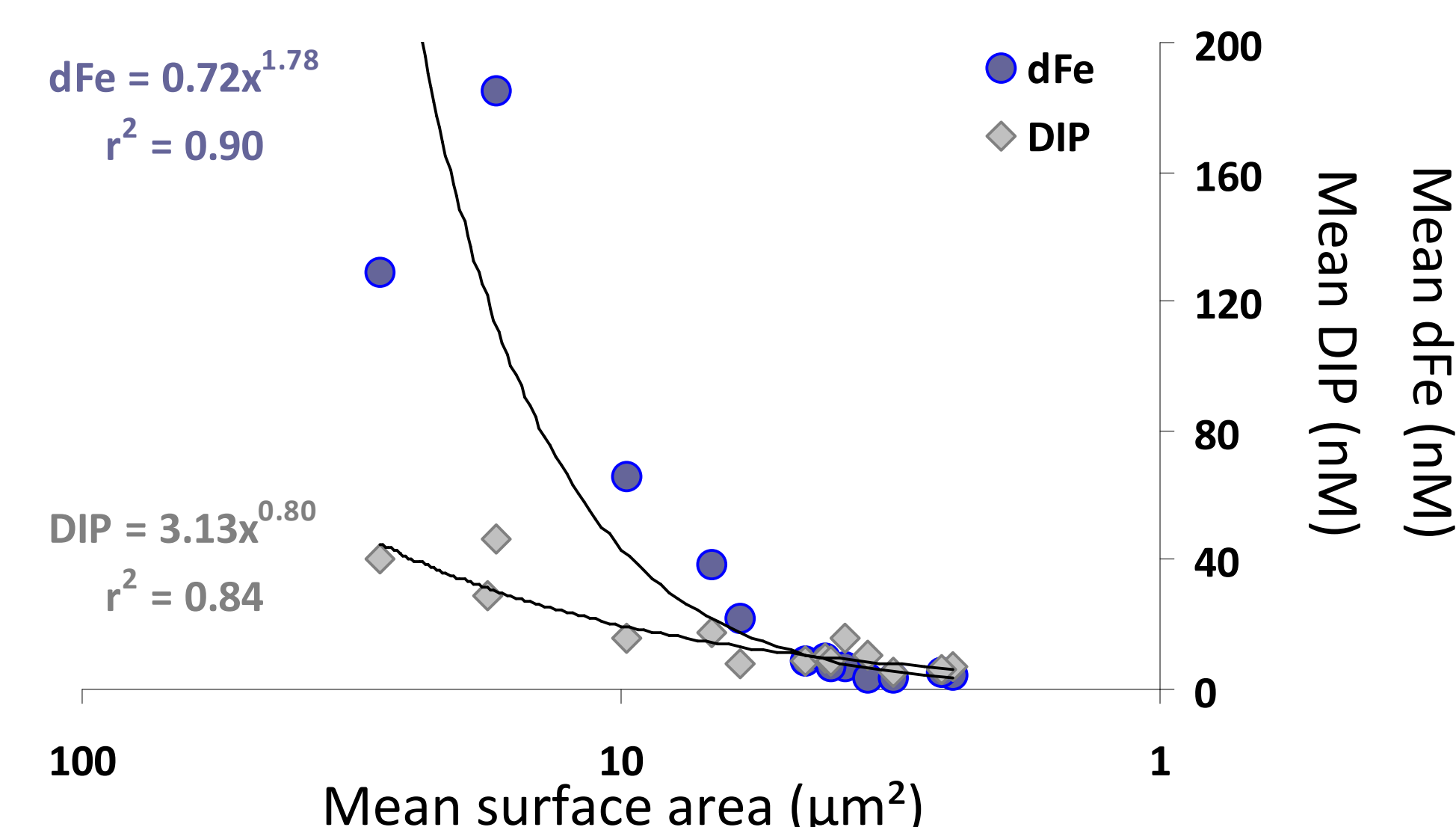
+ increase of CDOM

→ important role in initiating aggregation
→ accelerating the scavenging and the particulate export of atmospheric Fe and P

Rapid kinetic of formation of colloidal Fe and P (0.015 – 0.2 μm) after seeding but short residence time in surface (not shown)
→ preferential loss of colloidal Fe and P by transfer of the sorbed species from the colloidal pool to larger aggregates

February: Submicron particles active in Fe and P adsorption

Once equilibrium of the dissolution process reached
→ **Hyp:** submicron particles represent an important substrate for adsorption as they settle



Conclusions and Perspectives

- Consideration of short time scale very important for experimental evidence of dissolved-particulate exchanges
- Atmospheric nutrients inputs must be considered along with the settling of the atmospheric particles
- Dust deposition events may have different responses as a function of seawater characteristics

- Colloidal pumping plays a crucial role in the fate of atmospheric new nutrients
- Depending on *in situ* conditions (DOM), submicron lithogenic particles could have a “cleaning effect” for atmospheric nutrients
- Toward a better consideration of the dynamic and characterization of the colloidal pool (TEP – Size speciation of nutrients)